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CORSO DI LAUREA IN BIOTECNOLOGIE VEGETALI E MICROBICHE

Biomass Production by
***Nannochloropsis gaditana* CCMP527 in**
Outdoor:
Effects of Photobioreactor Type,
Dilution Rate and Solar Irradiance

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Abstract

In the industrial and development society, worldwide petroleum consumption has steady increased resulting in higher standards of living and development transportation and trucking. This will affect the stability of ecosystems and global climate as well as global petroleum reserves. Government programs for reduced reliance on fossil fuel, has involved specific research plan for fuel from renewable resources, like biodiesel. The production of biodiesel consists on a transesterification of saponifiable fatty acids with a short chain alcohol, to form acid alkyl (i.e. the biodiesel) and glycerol. The microalgae are considered the only alternative to current biodiesel crops, for a series of advantages like not require arable land and the use of herbicide and pesticide for their growth, and their cultivation can occur in brackish water, reducing the consume of freshwater. This study was aimed to evaluate the effects of dilution rate and some technical features, in three type of biophotoreactors (tubular, raceways and vertical flat panels) on biomass volumetric productivity ($\text{g L}^{-1} \text{d}^{-1}$). Also biochemical analysis of the biomass (lipid, fatty acids and proteins) were evaluated when the steady state of biomass production was reached. The study showed the optimal dilution rate was at 0.28d^{-1} in every photobioreactors and tubular shown the higher biomass volumetric productivity than the other photobioreactors. No significant correlation between the air quantity and sump at fix dilution rate were found, while in flat panels significant correlations between the distances from black panels were found. The biochemical data shown a higher content in lipids in tubulars than the other photobioreactors, with a prevalence of saturated and monounsaturated fatty acids than the polyunsaturated fatty acids in almost all photobioreactors. In conclusion, this study shown a superior performance of tubulars photobioreactors in biomass productivity and fatty acids contents for biodiesel. However, the costs of this system are very high and prohibitive to compete with other biodiesel sources like also petroleum derived diesel.

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Introduction

1.1 In search of the sustainable fuel

Petroleum is a natural product which derives by the anaerobic digestion of plant biomass in high condition of pressure and temperature. In the industrial society, worldwide oil consumption has steadily increased, resulting in higher standards of living, increased transportation and trucking and increased use of plastic and petrochemical (Bunnell, 2007). The transportation is the sector that used more oil, approximately 60 % of global oil consumption (International Energy Agency, 2012) and one-fifth of global carbon dioxide emissions are created by transportation sector (Goldemberg, 2008).

According to Matsuo et al. (2013), in 2010 were consumed 84 million barrels per day (Mbd^{-1}) of that reach 114 Mbd^{-1} on 2035, with a increased at an annual rate of 1.2 %. Around the world, there were about 806 million cars and light trucks on the road in 2007 (Plunkett, 2007), and these values are project to increased to 1.3 billion by 2030 and to over 2 billion vehicles by 2050 (World Business Council Sustainable Development, 2004). This growth will affect the stability of ecosystems and global climate as well as global oil reserves (Balat and Balat, 2009a,b).

Government programs to reduced reliance on fossil fuel (Afionis and Stringer, 2012; Koizumi, 2011; Koplow, 2006) has involved specific research plan for fuel from renewable sources, like biofuel.

The term bio-fuel is referred to as liquid or gaseous fuels for the transport sector that are predominantly produced from biomass. A variety of biofuels can be produced from biomass, including liquid fuels, such as biodiesel, bioethanol, methanol, and gaseous fuel, such as hydrogen and methane (Demirbas, 2008a). Biofuels can be classified based on their production technologies: first generation biofuels (FGBs), second generation biofuels (SGBs), third generation biofuels (TGBs) and fourth generation biofuels (Fatih Demirbas, 2009).

First generation biofuels (FGBs) refers to biofuel made from sugar, starch, vegetable oils or animal fats, using conventional technology. The basic feedstock for the production of FGBs are often grain, which can be fermented into bioethanol, or seed which can be pressed to yield vegetable oil for produce biodiesel (Thamsiroj and Murphy, 2009) Second generation biofuels (SGBs) is produced from feedstock of lig-

noncellulosic materials include cereal straw, forest residues, bagasse, purpose-grown energy crop such as vegetative grasses and short rotation forests like also organic waste like waste cooking oil and animal fats (Antizar-Ladislao and Turrion-Gomez, 2008; Fatih Demirbas, 2009). Third generation biofuels (TGBs) is based on algae or cyanobacteria that convert sunlight, water and carbon dioxide to algal biomass (Chisti, 2007). The fourth generation biofuel is the metabolic engineering of crops or microalgae, for enhance the production of biofuels and also the conversion of vegetal oil and biodiesel into biogasoline using most advanced technology (Demirbas, 2009; Lu et al., 2011).

While some researchers emphasized on the benefit of biofuels (Schubert, 2006; Somerville, 2006), others have criticized the economics and carbon mitigation of biofuel production (Palmer, 2007; Righelato and Spracklen, 2007; Waltz, 2007). Because of the competition with agriculture for arable land used for food production, the first generations biofuels systems shows many criticism. In fact, shift arable land use from food crops to energy crops leads to an increase in the food price and fall in the stocks of food products, with respective decline of exports (Rathmann et al., 2010). The increased pressure on arable land currently used for food production could also lead to severe food deficiency, in particular for developing world, where already more than 800 million people suffer from hunger and malnutrition (Schenk et al., 2008). Furthermore, plants are cultivated in an intensive way, in which many pesticides and fertilizer are used. This causes the contamination of surface water and problems like, eutrophication and eco-toxicity (Petrou and Pappis, 2009).

The increased criticism of the sustainability of many first generation biofuel, has raised attention to the potential of second generation biofuels. Second generation biofuels production has the potential to promote benefits such as: consuming waste residues, promote rural development and improved economic conditions in emerging and developing regions (Singh et al., 2011a). However second generations biofuels could become unsustainable if they compete with food crops for the available lands (Anselm, 2010) and the actual bioconversion technology (i.e., cellulose treatment, cellulose conversion and hydrolysis) need to be improved and optimized for a lower process cost (Antizar-Ladislao and Turrion-Gomez, 2008)

1.2 Potential of third generation biofuels

Microalgae are considered as the only alternative to current biofuel crops such as corn and soybean, for a series of advantages (Chisti, 2007). In contrast with the best oil-producing crops, microalgal biodiesel has the potential to displace completely oil-derived transport fuels, without impacting supplies of food and other agriculture product (Chisti, 2008). For their grown don't require arable land (Chisti, 2007; Hu et al., 2008; Singh et al., 2011b) and use of herbicides and pesticides, contributing to reduce the resourcing required for the production of food crops and the environmental impacts (Rawat et al., 2011; Rodolfi et al., 2009). Microalgae have faster growth rate than oil-crops and utilize large fraction of solar energy that is converted on chemical energy (Huber et al., 2006).

They grow in aqueous media, but need less water than terrestrial crops and the

cultivation can occur in brackish water or in high saline waters, reducing the consumed of freshwater sources (Dismukes et al., 2008; Huber et al., 2006; Searchinger et al., 2008)

Nutrients for microalgae cultivation, like nitrogen and phosphorus, can be obtained from waste water. Therefore there is a potential for treatment of organic effluent from the agri-food industry (Cantrell et al., 2008). Furthermore, they can produce valuable co-products such as protein and residual biomass, which may be used as feed or fertilizer (Spolaore et al., 2006). Anyway, the most important aspect of microalgae, is their potential to provide several types of renewable biofuels.

In fact, many microalgae are rich in oil. Levels of 20-50 % are quite common (Chisti, 2007) and some of them exceeds 80 % of the dry weight of algae biomass (Banerjee et al., 2002). Contrarily, agriculture oil crops such as soybean and oil palm yet used for produced biodiesel, have fewer amounts of oil, 5 % less of the total biomass (Chisti, 2008). Microalgae also synthesize other fuel products, such as biohydrogen (Razeghifard, 2013), ethanol (Deng and Coleman, 1999) and methane by anaerobic digestion of the algal biomass (Spolaore et al., 2006).

1.3 Biodiesel

1.3.1 Biodiesel Production

The production of biodiesel starts from the extraction of the triglycerides from oil plant-derived or microalgae (Chisti, 2007). The chemical conversion of triglycerides in biodiesel, also called transesterification, involves a reversible reaction with an alcohol to form acid alkyl esters, i.e. biodiesel, and glycerol (Balat and Balat, 2010). The alcohol employed in the transesterification are generally short chain alcohols such as methanol, ethanol, propanol and butanol (Speight, 2008). However, methanol is usually used to shift the reaction equilibrium to the right side and produce more methyl ester as proposed product (Demirbas, 2008b). The reactions take place at low temperature (338 K) at modest pressure (2 atm) (Balat and Balat, 2010) and the principle output at the end, are biodiesel (86%) and glycerol (Lucia et al., 2006).

The presence of water is a very important factor in the biodiesel production, because it could give rise to hydrolysis of some of the produced ester, with consequent soap formation. The soap reduces catalysis efficiency, increases the viscosity of the biodiesel, leads to gel formation and makes difficult the separation of glycerol (Refaat et al., 2008).

Transesterification reaction can be also catalyzed to improve the reaction rate of the biodiesel production and these processes include alkalis (Kotwal et al., 2009; Leung et al., 2010), acids (Li et al., 2010; Miao et al., 2009) or enzyme (Chen et al., 2009; Liu et al., 2009).

Alkali-catalytic transesterification

The alkali-catalytic transesterification involves catalysts like alkaline metal alkoxides and hydroxides, as well as sodium or potassium carbonate, like catalysts (Balat and Balat, 2010). Alkaline metal alkoxides like sodium methoxide for methanolysis, are

the most active catalyst. This catalyst even if is applied at low molecular concentration (0.5 mol%) they gives very high yields (>98%) in short reaction times (30 minutes)(Schuchardt et al., 1998). Alkaline metal hydroxides like potassium hydroxide and sodium hydroxide, are cheaper than metal alkoxides, but less active. Nevertheless, increasing the catalyst concentration to 1 or 2 mol%, these catalysts can give the same high conversions of oil, than the alkaline metal alkoxides (Schuchardt et al., 1998). However, sodium methoxide is the most widely used biodiesel catalyst, with over 60 % of industrial plants (Huber et al., 2006).

In general, alkali-catalyzed transesterification process are carried out at low temperature and pressure (333-338 K and 1.4-4.2 bar) (Lotero et al., 2006) in to three steps (Fangrui and Milford, 1999).

The first steps is a nucleophilic attack of the alkoxide at the carbonyl group of the triglyceride. This generates a tetrahedral intermediate. In the second steps, the rearrangement of the tetrahedral intermediate forms a alkyl ester and the corresponding anion of the diglyceride. Finally, in the last step the alkyl ester deprotonates the catalyst, which now is able to react with a second molecule of alcohol, starting a another catalytic cycle. Diglycerides and monoglycerides are converted by the same process to a mixture of alkyl ester and glycerol (Demirbas, 2005; Schuchardt et al., 1998).

The alkaline catalyst presents a important advantages, e.g. short reaction time and relatively low temperature(Singh et al., 2009), that made this process the most commonly used for the biodiesel production (Fukuda et al., 2001). The major problem of this catalytic process is the impossibility to convert oils that contain significant amounts of fatty acids, like low cost oils and animal fats, in biodiesel. This because the fatty acids reacts with the alkaline catalyst, produced a lot of soap (Furuta et al., 2004), that inhibit the separation of biodiesel, glycerin and wash water (Canakci and Van Gerpen, 2003).

Acid-catalytic Transesterification

Biodiesel produced by transesterification reaction can be catalyzed by sulfuric, phosphoric, hydrochloric and organic sulfonic acids. Currently the acid-catalytic more used in biodiesel production are the organic acids, such as derivatives of toluene-sulfonic acids and, more often, mineral acids such as sulphuric acid (Cardoso et al., 2008).

Reaction conditions reported in the literature using the acid-catalyzed reaction, are summarized by Balat and Balat (2010). The mechanism of the acid-catalyzed transesterification, starts with a protonation of carbonyl group of the ester, that leads to the formation of carbocation. This latter, undergoes to a nucleophilic attack of the alcohol and produces a tetrahedral intermediate. This intermediate eliminates glycerol to form a new ester and to regenerate the catalyst (Meher et al., 2006). Major advantages of the use of this catalytic transesterification methods are the high yields and the possibility to do the transesterification reaction also if high water and fatty acids are presents in oil (Lotero et al., 2006; Oliveira et al., 2005). Despite this, acid-catalyzed transesterification has been largely ignored, mainly because of its relatively slower reaction rate, typically 4000 times slower than alkali catalysis

(Balat and Balat, 2010; Zhang et al., 2003).

Enzyme-catalyzed transesterification

Enzymatic catalyst like lipase (E.C. 3.1.1.3.) is able to make effectively the transesterification of triglycerides with short chain alcohol, in either aqueous or non-aqueous systems (Fukuda et al., 2001; Helwani et al., 2009; Korman et al., 2013; Meher et al., 2006). Enzyme-catalyzed reaction has the advantages to overcome several drawbacks reported for alkaline catalysis process like it energy intensive (enzyme-catalyzed required moderate temperature of 303 to 313 K), the difficult recovery of glycerol, the elimination of the catalyst from the products, high level of fatty acids and water could interfere with the reaction (Balat and Balat, 2010). Moreover, the lipase-catalyst can be easily separated from the glycerol product. In addition glycerol without contamination has a further value-added, lowering the net cost of biodiesel production (Balat, 2009; Fukuda et al., 2001; Parawira, 2009; Sotoft et al., 2010). However some important drawback to use lipase, limit the commercial use in biodiesel process. These include the high cost of the catalyst, the slow reaction rate, from 4 to 40 hours or more, and the alcohol-induced inactivation of the enzyme by high concentration of methanol used for biodiesel synthesis (Balat, 2009; Biermann et al., 2011; Fjerbaek et al., 2009; Sotoft et al., 2010).

Although the enzyme-catalyzed transesterification process are not yet commercially, recent advance in enzyme technology, such as the use of solvent-tolerant lipases and immobilized lipases, optimizing the reaction condition (temperature, solvent, pH, alcohol/oil molar ratio) and using of microbial lipases enzyme, have been studied in order to develop suitable characteristic for an industrial applications (Balat and Balat, 2010; Joshi and Vinay, 2007; Korman et al., 2013; Tan et al., 2010).

1.3.2 Technical Features of Biodiesel

Most biodiesel fuels are composed of only few different compounds and thus their properties are largely influenced by the fatty acids composition of the fuel. Affected physical properties may include: ignition quality, heat of combustion, cold flow properties, oxidative stability, viscosity and lubricity (Knothe, 2005). In turn the physical properties of the fuel can influence the exhaust emission and performance of the engine (Knothe, 2008).

Ignition Quality: The Cetane Number

As indicator of ignition quality, the Cetane Number (CN) is a prime indicator of fuel quality within of diesel engine. The CN of a diesel fuel is related to the ignition delay time, i.e the time between injection of the fuel into the cylinder and starting of injection (Knothe, 2005). An important consideration about the ignition quality, is influence of the CN by the structure of fatty acid methyl ester component (Bangboye et al., 2008). CN increases with an increasing of chain length and increasing saturation. Branched and aromatic compound have low CNs. Thus, compounds found in biodiesel, like methyl palmitate and methyl stearate have high CNs, while methyl linolenate has very low CN (Knothe, 2008).

Heat of Combustion

An important parameter for the quality of fuel is its energy densities, expressed as the heat of combustion (kJ g^{-1}) of the fuel. Fuel with a higher heat of combustion can accomplish more work than an equal amount of lower heat of combustion fuels (Knothe, 2005; Wahlen et al., 2013). In other words, the heat of combustion is important for estimating fuel consumption: the greater the heat of combustion, the lower the fuel consumption (Knothe, 2008). Like the CN number, the heat of combustion is influenced by the fatty acid methyl ester component. In fact, the heat of combustion increases with an the chain length and decreases with the unsaturation level (Knothe, 2008).

Cold Flow Proprieties

One of the major problem associated with the use of biodiesel is poor low temperature flow proprieties, indicated by relatively high cloud points (CP), and pour points (PP) (Knothe, 2005). The CP is the first solid become visible when cooling a diesel fuel; the PP is the temperature at which the fuel ceases to flow (Knothe, 2008).

However, when the CP temperature is reached (that is generally at higher temperature than PP) the fuel biodiesel stars to be cloudy due to the formation of crystals and solidification of saturates. Solid and crystal grow rapidly and agglomerate, clogging fuels line and filters causing major operability problems (Knothe, 2005).

Therefore, saturated fatty compound have significantly higher melting points that unsaturated fatty compounds. Thus biodiesel fuels derived from fats or oil with significant amount of saturated fatty compound will display higher CPs and PPS (Knothe, 2005).

Oxidative Stability

Several factor can affects the stability of biodiesel primarily during extended storage. Generally, factors such as presence of air, elevated temperatures or presence of metals, facilitate the oxidations (Knothe and Dunn, 2003). The reason for autoxidation is the presence of double bonds in the chain of many fatty acids. The autoxidation of unsaturated fatty compounds proceeds with different rates depending on the number ad position of double bonds (Frankel, 1998). The position allylic to double bonds are especially susceptible to oxidation. This true especially for polyunsaturated fatty acids in which the *bis*-allylic position is common. Fatty acids such as linoleic acid (double bonds at $\Delta 8$ and $\Delta 12$, giving one *bis*-allylic position at C-11) and linolenic acid (double bonds at $\Delta 9$, $\Delta 12$ and $\Delta 15$ giving two *bis*-allylic position at C-11 and C-14) are very prone to autoxidation (Frankel, 1998). This biodiesel characteristic is important because mos of biodiesel contain significant amount of esters of oleic, linoleic or linolenic acids, which influenced the oxidative stability of the fuels.

Viscosity

Viscosity of biodisel is the most important physical parameter, since it affects the operation of fuel injection equipment, particularly at low temperature when the

increase in viscosity, affects the fluidity of the fuels (Balat, 2008). Viscosity increases with chain length and with increasing degree of saturation, like also configuration of double bond of the *cis* double bond configuration giving a lower viscosity than *trans* (Knothe, 2005).

Lubricity

Increasing strict regulation on the sulphur content of commercial diesel fuels results in the decrease in its lubricity. This reduced lubricity could be deleterious to the engine and fuel injection. Vegetable oil based diesel fuel additives may be a possible solution to this problem (Geller and Goodrum, 2004). In fact, fatty acids possess excellent lubrication propriety and unsaturated acids exhibited better lubricity than saturated species (Kenesei and Ecker, 2003). Adding biodiesel at low levels (1-2 %) restores the lubricity to low-sulphur oil-derived diesel (Knothe, 2005). Like other physical parameter described before, also the structure of fatty acid methyl ester component influences the lubricity. In particular, ethyl ester had improves lubricity compared to methyl ester (Waynick, 1997).

Exhaust Emissions

The combustion of petroleum-based diesel fuel is a major source of GHG emissions. Apart from these emissions, petroleum-based fuel is also the major source of other air contaminants including carbon monoxide, nitrogen oxides, sulphur oxide, particulate matter (PM) and volatile organic compound (VOCs) (Chien et al., 2009). Biodiesel shows reduced exhaust emissions compared to petroleum diesel; many studies have concluded that biodiesel use results in the reduction of unburned hydrocarbon, particulate, carbon monoxide emissions (Lackey and Paulson, 2012). In contrast to these improvements in emissions for biodiesel, most emissions studies have found that biodiesel produces more nitrogen oxides emissions, than petroleum diesel (Lapuerta et al., 2010). McCormick et al. (2001) and Knothe et al. (2006) found that nitrogen oxides emissions increased with increasing amounts of unsaturated fatty acids as well as the chain length (>C16) of this latter. On the other hand, additional study have found that oils containing less polyunsaturated fatty acids, such as palm oil and coconut oil, have lower nitrogen oxides emissions (Benjumea et al., 2011; Bugosh et al., 2011; Cecile et al., 2012; Lin et al., 2008).

1.4 Oleaginous Microalgae

The algae are a polyphyletic, non-cohesive and artificial assemblage of photosynthetic organisms, that includes microalgae and other highly diverse groups of micro-organism know as microalgae. Most algae are aquatic in either fresh or marine waters, but some species may be found growing in such diverse habitats as tree trunks, snow bunks, hot springs or even within desert rocks. They can be motile and unicellular or pluricellular, and non-motile at some stage of their life (Gualtieri, 2000). The number of algal species has been estimated in the range from one to ten million,

and most of them are microalgae. This assemblage of organisms includes prokaryotic and eukaryotic algae. (Norton et al., 1996). Prokaryotic members are grouped in two divisions: *Cyanophyta* (blue-green algae) and *Prochlorophyta*, whereas eukaryotic members are grouped into nine divisions: *Glaucophyta*, *Rhodophyta* (red algae), *Heterokontophyta*, *Haptophyta*, *Cryptophyta*, *Dinophyta*, *Euglenophyta*, *Chlorarachniophyta* and *Chlorophyta* (green algae) (Barsanti and Gualtieri, 2005).

The total oil and fat content of microalgae ranges from 1% to 70% of the dry weight and tend to be inversely proportional to the rate of growth with greater accumulation during stationary phase (Borowitzka, 1988).

The most commonly synthesized fatty acids have chain lengths that range from C16 to C18, similar to those of higher plants (Ohlrogge et al., 1995), and in general saturated and mono-unsaturated fatty acids are predominant in most algae examined (Borowitzka, 1988). However the fatty acid profile is also influenced by ageing of the culture, growth stage, environmental conditions (pH and temperature), nutrient availability and light intensity (Hu et al., 2008; Metting F.B., 1996).

1.4.1 Biosynthesis of Lipids in Microalgae

Microalgae can fix carbon dioxide into sugar using energy from the sun. The fixed sugars are further processed to produce acetyl-CoA (Coenzyme A), and more than one pathway may contribute to maintain the acetyl-CoA pool. Acetyl-CoA provided by photosynthesis serves as the precursor for fatty acid synthesis in the chloroplast. Fatty acids are the building block of many cellular lipid types including triacylglycerol. The first committed step of fatty acid synthesis is catalyzed by a multifunctional enzyme complex, i.e. acetyl CoA carboxylase (ACCase) which produces malonyl-CoA from acetyl CoA and bicarbonate. ACCase has been cloned and characterized from two eukaryotic microalgal species. Before being further used by the fatty acid synthase machinery, the malonyl group is transferred from CoA to ACP (acetyl carrier protein) catalyzed by a malonyl-CoA:acyl carrier protein malonyltransferase. The common 16- or 18-carbon fatty acids are formed by a series of two-carbon chain elongating reactions catalyzed by a multi-subunit enzyme named fatty acid synthase (FAS). Fatty acid synthesis requires stoichiometric amount of ATP, acetyl-CoA and NADPH for each two carbon added to the growing acyl chain. Photosynthetic reactions are thus essential not only in providing a carbon source but also in generating reducing power (NADH and NADPH) and energy (ATP). For most algal species, the final acyl chains emerging from the chloroplast are 16- or 18- carbons in length. The chain-elongating reaction is terminated by the action of acyl-ACP thioesterase (FAT). The specificity of this enzyme usually determines the final chain length of the product. The released free fatty acids can cross the plastid envelope membrane where they are esterified to CoA via another enzymatic reaction catalyzed by long chain acyl-CoA synthase (LACS).

The Biosynthesis of Triglycerides in Microalgae

Like other higher plants and animals, microalgae are able to biosynthesize triglycerides to store substance energy. Generally, L- α -phosphoglycerol and acetyl-CoA

are two major primers in the biosynthesis of triglycerides. The L- α -phosphoglycerol mainly derives from phosphodihydroxyacetone which is the product of glycolysis process. One of the hydroxyl in L- α -phosphoglycerol reacts with there acetyl-coA to form Lysophosphatic acid and this combines with another acetyl-CoA to form phosphatidic acids. This two reactions are catalyzed by glycerol phosphate acyl-transferase. In the following steps, lysophosphatidic acid is hydolyzed by phosphatidate phosphatase to form diglyceride which is combined then with the third acetyl-CoA to complete the biosynthesis of triglycerides. The last reaction is catalyzed by glyceryl diester transacylase (Huang et al., 2010).

1.5 The Phylum Heterokontophyta

Heterokont alage are monophyletic group that includes all photosynthetic organism with tripartite tubular hair on the mature flagellum as well as some non-photosynthetic relatives. Sometimes, they have secondarily reduced or lost tripartite hairs (Andersen, 2004).

The new technologies availability, like transmission electron microscopy and new biochemical and cladistic studies approach, permit the description of many new taxa in *Heterokont*, including several classes. Currently are recognized 15 classes: *Bacillariophyta*, *Bolidophyceae*, *Chrysomerophyceae*, *Chrysophyceae*, *Dictyochophyceae*, *Eustigmatophyceae*, *Pelagophyceae*, *Phaeophyceae*, *Phaeothmniophyceae*, *Pinguiphyceae*, *Raphidophyceae*, *Schizocladophyceae*, *Synurophyceae* and *Xanthophyceae* (Andersen, 2004).

Ecology

Heterokont algae widely distribute but the varies classes can colonize different environment. *Bolidophyceae*, *Chrysomerophyceae*, *Pelagophyceae*, *Pinguiphyceae* and *Schizocladophyceae* are only know from marine environment (Andersen and Preisig, 2002a; Billard, 1984; Guillou et al., 1999b; Kawachi et al., 2002a; Kawai et al., 2003). *Phaeophyceae* are almost exclusively marine organism, but five freshwater genre are known (Bond and Wynne, 1985). *Synurophyceae* are probably restricted to freshwater, although a couple of dubious marine occurrences have been reported (Andersen and Preisig, 2002b). *Chrysophyceae*, *Phaeothamniphyceae* and *Xanthophyceae* are predominately freshwater organism, although a substantial number of *xanthophytes* are terrestrial (Andersen, 2004; Kristiansen et al., 2001; Preisig and Hibberd, 1983; Starmach, 1985). *Dictyochophyceae* occur in both marine and freshwater habitats (Moestrup, 1995; Moestrup et al., 2002), and *Eustigmatophyceae* occur in freshwater, marine and terrestrial habitats (Hibberd, 1990a). In *Rhaphidophyceae* there is one group with fucoxanthin-violaxanthin type pigment that lives in marine environment, and one with heteroxanthin-diatoxanthin type pigments that lives in freshwater (Heywood, 1990; Heywood and Leedale, 2002; Potter et al., 1997). Finally, diatoms are found in all common habitats supporting life (Round et al., 1990).

1.5.1 Cell Biology

Chloroplast and pigments

The chloroplast structure of all heterokont algae share some features (Andersen, 2004). The chloroplast is surrounded by the chloroplast endoplasmic reticulum, and thus four membrane separate the stroma from the cytosol. Each chloroplast lamella consists of three adpressed thylacoids. Most *Heterokont* classes, for exception of *Eustigmatophyceae* excepted, have a girdle lamella which is a saclike three-thylacoids structure that encloses all other lamellae. Besides, in most heterokont classes the outer membrane of the chloroplast endoplasmic reticulum is continuous with the outer membrane of the nucleus. The inner chloroplast endoplasmic reticulum is considered to be the remnant plasmalemma of an ancient endosymbiotic event or derived from the outer nuclear envelope as well (Andersen, 2004). Many *Heterokont* swimming cells as well as some *Pavlovophyceae* have an eyespot that is located within the chloroplast or associated with in (Andersen, 2004; Green, 1980). Eyespot are part of photoreceptors apparatus (called the eyespot apparatus), shielding light so that the other elements can more precisely determine the direction of light (Foster and Smyth, 1980). Chloroplast function primarily for photosynthesis, and *Heterokont* have a wide variety of light-harvesting pigment, many of which are photosynthetically active. All *Heterokont*, except *Eustigmatophyceae*, have one or more types of chlorophylls c and they are rich in carotenoids, giving them a golden or brown colour (*Eustigmatophyceae* and *Xanthophyceae*). In addition to other roles (e.g. ultra-violet light protection, photosynthetic quenching), one or more photosynthetically active carotenoids are usually present (Alberte and Andersen, 1986; Andersen, 2004; Porra et al., 1997). Finally, not all species have chloroplast. Leucoplasts (unpigmented plastids) are present in some *Chrysophytes* (*Paraphysomonas* and *Spumella*) and in *Dictyochophyceae* (*Pteridomonas* and *Ciliophyris*) (Andersen, 2004; Preisig and Hibberd, 1983).

Cell coverings

Heterokont algae have a wide range of cell covering. *Bolidophytes* are naked flagellates (Guillou et al., 1999b), diatoms have siliceous frustules (Round et al., 1990); *Chrysomerophytes* have cell wall; *Chrysophytes* have cell wall, organic loricas, organic or silica scale cases, gelatinous covering and completely naked cell (Kristiansen et al., 2001; Preisig and Hibberd, 1983; Starmach, 1985); *Dictyochophytes* have silica skeletons, organic scales or naked cells (Moestrup, 1995; Moestrup et al., 2002); *Eustigmatophytes* have cell wall (Hibberd, 1990a); *Pelagophytes* have cell wall, thecae, gelatinous covering, naked cells (Andersen and Preisig, 2002a). *Phaeophytes* have cellulosic cell walls impregnated with alginates and often interconnected via plasmodesmata (Andersen, 2004; Pueschel and Stein, 1983); *Phaeothamniophytes* have cell wall (Craig Bailey et al., 1998); *Pinguiophytes* have mineralized loricas, gelatinous coverings, or naked cells (Kawachi et al., 2002a,b,c); *Raphidophytes* are naked cells (Heywood, 1990; Heywood and Leedale, 2002); *Schizocladophyceae* has cell walls without cellulose but impregnated with alginate (Kawai et al., 2003); *Syrinophytes* have bilaterally symmetrical silical scales glued together to form a highly or-

ganized scale case (Ludwig et al., 1996); *Xanthophytes* have predominately cell wall as plasmodial and naked forms (Hibberd, 1990b).

Flagellar apparatus

The typical swimming cell of *Heterokont* algae has two flagella, a long immature flagellum and a short mature flagellum (Andersen, 2004). Flagellated vegetative cells of *Bolidophyceae*, *Chrysophyceae* and *Raphidophyceae* as well as most vegetative cells of *Synurophyceae* and *Phaeomonas* (*Pinguiphyceae*) have two typical flagella (Andersen, 1989, 2004; Guillou et al., 1999a; Heywood, 1990; Honda and Inouye, 2002). Similarly, also flagellated zoospores or sperm of *Chrysomerophyceae*, *Eustigmatophyceae*, *Phaeophyceae*, *Phaeothamniophyceae*, *Schizocladophyceae* and *Xanthophyceae* as well as some *Pelagophyceae* have two typical flagella (Andersen et al., 1998; Billard, 1984; Hibberd, 1990a; Kawachi et al., 2002b; Lobban et al., 1995; O’Kelly, 1989).

Stages of development of flagellum in *Heterokonts* are unknown, but it may be similar to that for green algae. An immature flagellum is produced de novo during cell division, and the previous immature flagellum is transformed into mature flagellum by a process termed flagellar transformation (Wetherbee et al., 1988).

Typical *Heterokont* flagellum has tripartite tubular hairs arranged in two rows along the immature flagellum. The flagellum beat is sinusoidal; the hairs reverse the thrust of the flagellum, and therefore, the beating flagellum pulls the cell forward (Andersen, 2004). Members of *Chrysophyceae* and *Synurophyceae* have lateral fibres on the central shaft of the tripartite hair, but such lateral hairs are absent in all other *heterokont* algae (Andersen, 1989, 2004). On other hand, there are no tripartite hairs on the emergent flagellum of flagellated eggs of *Laminaria angustosa* Kjellman (Motomura and Sakai, 1988), or the zoospores of *Glossomastix* and *Polypodochrysis* (*Pinguiphyceae*) (Kawachi et al., 2002d; O’Kelly, 2002). Flagellum are anchored in the cell with various structure that are generally referred to as the flagellar root apparatus. For a description of the all roots parts of the flagellum, it can be refer to Andersen (1992).

Other Ultrastructural Features

All *Heterokonts* have mitochondria with tubular cristae (Andersen, 2004; Stewart and Mattox, 1980), *Heterokont* algae have typical Golgi bodies, and in most classes (*Dictyochophyceae* excepted), Golgi bodies are anterior to the nucleus, with cisternae adjacent the nuclear envelop (Andersen, 2004). Mucocysts are common in *Raphidophyceae* (Heywood, 1990; Heywood and Leedale, 2002) and various mucosal vesicles occur in some members of *Chrysomerophyceae* and *Chrysophyceae* (Andersen, 1982; Hibberd, 1970; Mignot, 1977).

Mitosis

Mitosis is known only for a few *Heterokont* algae. In diatoms and most *Chrysophyceae* the nuclear envelope disperses during prophase. Spindle microtubules

attach to either basal bodies (diatoms) or the striated flagellar roots (*Chrysophyceae*) (Andersen, 1989, 2004; Bouck and Brown, 1973; Slankis and Gibbs, 1972; Tippit et al., 1980). However, in *Hydrurus* (*Chrysophyceae*), the nuclear envelope remains largely intact, with openings at the poles (Vesk et al., 1984). *Pelagococcus* (Vesk and Jeffrey, 1987), *Synura* (Andersen, 1989) and most *Phaeophyceae* (Green, 1980), behave similarly to *Hydrurus*. *Vaucheria* (*Xanthophyceae*) has an intact nuclear envelope at metaphase, and spindle microtubules from completely within the nuclear envelope (Ott and Brown Jr, 1972). *Vacuolaria* (*Raphidophyta*) is perhaps the most unusual situation, in which the nuclear envelope of daughter cells forms inside the dispersing old mother nuclear envelope heywood,heywood2. Mitosis has not been reported for *Bolidophyceae*, *Chrysomerophyceae*, *Dictyochophyceae*, *Eustigmatophyceae*, *Phaeothamniophyceae*, *Pinguiphyceae* and *Schizocladophyceae* (Andersen, 2004).

1.6 The Class Eustigmatophyceae

The class of *Eustigmatophyceae* includes unicellular and coccoid organism with a polysaccharidic cell wall. Its name derived from a particular characteristic of this class: the presence of a eyespot adjacent to flagellum and not to the chloroplast (Kawai and Kreimer, 2000). Generally, they living in freshwaters and soil and produce a small number of zoospores with usually two or one apical flagella. The chloroplasts have only chlorophyll a and violaxanthin is the major light-harvesting carotenoids pigment. The microalgae of this class are a promising source of eicosapentaenoic acid (20:5n3) that is content in citosol and photosynthetic lamella (Cohen, 1999).

1.7 Nannochloropsis gaditana

Nannochloropsis sp. is a picoplankton genus of marine environment. The cells are small (2-4 μm in diameter), spherical to slightly ovoid, non-flagellate, with a polysaccharide cell wall. They have one single chloroplast without pyrenoid and containing several bands of photosynthetic lamellae, each with three thylakoids per band (Boussiba et al., 1987).

Phylogenetic Relationships

A recent study of Radakovits et al. (2012) about the sequencing and assembling of nuclear and organellar genome of *Nannochloropsis gaditana* CCMP526, allowed to performed a phylogenetic analysis of the *N.gaditana*. *N. gaditana* present gene homologue with brown algae and the pelagophyte *Aureococcus anophagefferens*, green algae *Chlorella variabilis* and *Chlorella reinhardtii*, red algae *Cyanidioschyzum merolae* and diatoms *Thalassiosira pseudonana* and *Phaeodactylum tricornutum*. This analysis confirms the close evolutionary proximity between the *Eustigmatophyceae* and *Phaeophyceae*. Instead, the homologue between the other species of *Nannochloropsis* reported the that *N. gaditana* is most closely related to *N. salina*.

1.8 Microalgal Growth

The growth of microalgae in a bioreactor, where the nutrients are available and the environmental conditions are suitable, presents a classical microbial growth trend with four phases: adoption (lag phase), exponential (log phase), steady state, log death (Becker, 1993).

The culture of microalgae may be performed by two modes: the batch mode and continuous mode. As reported by Richmond (2008), in a simple batch culture systems, a limited amount of complete culture medium and algal inoculum are placed in a culture vessel and incubated in a favourable environment for growth. At this stage, the microalga starts to grow following the different growth phases described before. In a continuous flow culture, fresh culture medium is supplied to the homogeneously mixed culture, and culture is removed continuously or intermittently. The approach is based on the observation that substrates are depleted and products accumulate during growth. Eventually, culture growth ceases due to depletion of the growth limiting substrate or accumulation of a growth-inhibiting product. To sustain the cell growth, the growth-limiting substrate needs to be replenished and the growth inhibitory products need to be removed or diluted by adding fresh. In the following section, we will introduce the theory of the principle of continuous culture elaborated by (Herbert et al., 1956; Richmond, 2008). For deep information and mathematical demonstration about D_c and D_m (see below), it is recommended to read Herbert et al. (1956).

1.8.1 Principles of Continuous Culture

For simplicity, in a bioreactors the medium feed rate and the rate of removal of culture (F) are the same and the culture volume is constant (V), and two peristaltic pumps are available for delivery of the medium and for removal of the latter. The increase in biomass in the culture can be expressed as follows:

$$\text{Net increased of biomass} = \text{Growth} - \text{Output}$$

For an infinitely small time interval dt , this balance for the culture could be written as:

$$\frac{dx}{dt} = \mu x - Dx$$

where dx is the increase in biomass concentration (g L^{-1}), dt infinitely small time interval (hour), μ is the specific growth rate (h^{-1}), x is the biomass concentration (g L^{-1}), and D is the dilution rate (h^{-1} or d^{-1}), that is the rapport between the rate of removal and the culture volume (F/V). For example, medium is added into and culture removed from a 5 L algal culture, at a flow rate of 10 L h^{-1} . The rate of dilution of the culture is $10/5 = 2 \text{ h}^{-1}$. That is, the culture is diluted two times every hour.

So, if $\mu > D$, dx/dt is positive and the concentration of the organism will increase, while if $D > \mu$, dx/dt is negative and the concentration of the organism will decrease, eventually to zero. When $\mu = D$, $dx/dt = 0$ and x is constant; i.e.

we have a steady state in which the concentration of organism does not change with the time.

Theory indicates that is possible to fix the specific growth rate of an algal culture at any value from zero to maximum, by adjusting the dilution rate of the culture. So, is possible to determinate the dilution rate in which correspond to the maximal growth rate (μ_{max}) of a culture (also called the critical value of dilution, D_c) and the dilution which the product Dx is a maximum (indicated D_m), i.e. the dilution rate in which gives the maximum output of organism in unit time.

1.8.2 Chemostat

The special type of continuous culture where the rate of addition of medium and culture volume is thus maintained at constant level, is called chemostat (*constant chemical environmental*). Chemostat is widely used in research, for it allows full adjustment of the cell's physiology to the prevailing culture condition and maintaining the specific growth rate at pre-determined values (Iehana, 1983; Lee and Soh, 1991; Molina Grima et al., 1994b; Pirt et al., 1980). Culture parameter such as temperature, pH and substrate concentration can readily adjust and study at fixed specific growth rates. While in a simple batch culture, a change in culture parameters leads inevitably to altered specific growth rate. Such a batch culture could not differentiate between the effects of culture parameters and the specific growth rate.

1.9 The Photobioreactors

Photobioreactors (PBR) are reactors in which phototrops (microbial, algal or plant cells) are grown or used to carry out a photobiological reaction (Tredici, 2002). Two systems have been deployed and are based on open pound and closed photobioreactor technologies (Borowitzka, 1999). The technical viability of each systems is influenced by intrinsic proprieties of the selected algae strain used, as well as climatic condition and the cost of land and water (Borowitzka, 1995).

1.9.1 Enviromental Factors

The most relevant environmental factors that affects the growth of microalgae in a photobioreactor includes light, temperature, pH, salinity, dissolved oxygen (DO). The growth of microalgae is influenced also by biological factors that might constrain microalgal growth rates include predation, viruses, competition and growth of epiphytes (Carlsson and Bowles, 2007). Finally, microalgal growth can be affected by such reactors operating conditions as hydraulic residence time, harvesting rates, gas transfer and mixing equipment, that affect carbon dioxide availability and light exposure (Kumar et al., 2010a).

Light

Sunlight is the most common source of energy for microalgae. Light intensity requirement relatively low compared with higher plant. Microalgal activity usually

risks with increasing light intensity up to $400 \mu\text{mol m}^{-2} \text{s}^{-1}$ (Munoz and Guieysse, 2006). However, above a certain value of light intensity, a further increase in light level actually reduces the biomass growth rate. This phenomenon is called photoinhibition and results from generally reversible damage to the photosynthetic apparatus as a consequence of excessive light (Rubio et al., 2003). For example, Molina Grima et al. (1996), reported in *Isochrysis galbana* a photoinhibition for irradiance above $1630 \mu\text{mol m}^{-2} \text{s}^{-1}$. So, it is important during a culture growth process, especially in outdoor, know if the culture is photoinhibited or not in relation to obtain maximal biomass productivity. Chlorophyll fluorescence can be exploited to evaluate the photochemical status of a cell. The ratio between the variable fluorescence, i.e. the difference between the maximum fluorescence F_m (all Q_a reduced) and minimum fluorescence F_0 (all Q_a oxidized), and the maximum fluorescence is a convenient measure of the potential maximum quantum yield of PSII, and it has been assumed as an index of photoinhibition (Richmond, 2008). The F_v/F_m ratio is a convenient measure of the potential maximal quantum yield of PSII, and it has been assumed as an index of photoinhibition (Björkman and Demmig, 1987). Decrease in the F_v/F_m ratio has been found correlated to reduction in the quantum yield of oxygen evolution or carbon dioxide uptake (Genty et al., 1989). Another important factor that influences the light availability for the cells inside the photobioreactors, is the transparency of the external material of this latter. In fact, the light scattering effects may lead to a misvaluation of the absorbed flux (Aiba and Ogawa, 1983). However mathematical models to determine the average irradiance (I_{av}) inside the vessel, were proposed by Molina Grima et al. (1997). Measuring the incident irradiance on the vessels is possible to determine the average irradiance inside the vessel that is available for the culture. Finally, also the orientation of the photobioreactors is important for light availability. The solar radiation on a fixed surface can be estimated as a function of its location and position (Liu et al., 2009).

Temperature

Temperature is one of the major factors that regulates cellular, morphological and physiological response of microalgae and low temperature leads to inhibition of microalgal growth (Munoz and Guieysse, 2006). The optimal temperature varies among microalgal species. Optimal growth temperature of 15-26 °C have been reported for some microalgal species (Ono and Cuello, 2003). Daytime higher temperatures were observed to have clearly favourable effects on microalgal growth rates due to photosynthesis, except when the night temperature was as low as 7 °C (Kumar et al., 2010a).

pH

Most algae species are favoured by neutral pH whereas some species are tolerant to higher pH e.g. *Spirulina platensis* at pH 9 (Hu et al., 2008), or lower pH e.g. *Chlorococcum littorale* at pH 4 (Kodama et al., 1993). There is a complex relationship between carbon dioxide concentration and pH in microalgal bioreactors systems. Increasing carbon dioxide concentrations can lead to higher biomass productivity, but also decreases pH, which can have an adverse effect upon microalgal physiology.

By contrast microalgae have been shown to cause a rise in pH to 10-11 in open ponds because of carbon dioxide uptake (Oswald, 1988). This increase in pH can be beneficial for inactivation of pathogens in microalgal waste waters treatment, but can also inhibit growth (Kumar et al., 2010a).

Dissolved Oxygen

Excessive dissolved oxygen, OD (i.e. $>35 \text{ mg L}^{-1}$) can inhibit the metabolic process. The OD supersaturation in enclosed bioreactors designed for mass microalgal cultivation can reach levels as high as 400%, thus severely inhibits microalgal growth (Lee and Lee, 2003). Furthermore, microalgae are negatively charged on their surface, so they can strongly adsorb polyvalent cations like heavy metal. This latter could inhibit the microalgal photosynthesis because they can replace or block the prosthetic metal atoms in the active site of relevant enzymes, or induced morphological changes in the microalgal cells (Kumar et al., 2010a; Munoz and Guieysse, 2006).

Carbon Dioxide Uptake

Usual sources of carbon dioxide for microalgae include: atmospheric carbon dioxide, carbon dioxide from industrial exhaust gases and carbon dioxide chemically fixed in the form of soluble carbonates (e.g. sodium bicarbonate and sodium carbonate). Atmospheric carbon dioxide levels (0.0387% (v/v)) are not sufficient to support the high microalgal growth rates and productivities needed for full-scale biodiesel production. Waste gases from combustion process, however, typically contain $>15\%$ (v/v) of carbon dioxide, in which is sufficient for large-scale production of microalgae (Lackner, 2003). Flue gases that contain carbon dioxide at concentration ranging from 5 to 15% of (v/v) have indeed been introduced directly into ponds and bioreactors of various configuration that contain several microalgal species (Kumar et al., 2010a).

Nutrient Requirement

Apart for carbon, nitrogen is the most important element that is required for microalgal cultivation (Becker, 1993) and, as constituent of both nucleic acids and proteins, nitrogen is directly associated with primary metabolism of microalgae. Fast-growing microalgae species prefer ammonium rather than nitrate as primary nitrogen source (Jin et al., 2006). Under partial nitrogen deprivation, microalgae grow at lower rates, but produce significantly more lipids, which are reserve compounds synthesized under stress condition, even at expense of lower productivities (Lardon et al., 2009). Phosphorus is the third most important nutrient for microalgal growth and should be supplied to a significant excess as phosphate because not all phosphorous compounds are bioavailable, being combined with metal ions (Kumar et al., 2010b). Seawater with commercial nitrate and phosphate fertilizer is commonly used for production of microalgae (Green, 1980), whereas trace of metal (magnesium, calcium, manganese, zinc, copper and molybdenum) and vitamins are typically added to the cultivation (Becker, 1993).

1.10 The Photobioreactors Systems

1.10.1 Open Pound

Algae cultivation in open pound production systems has been used since the 1950s. These systems can be categorised into natural waters (lakes lagoon and ponds) and artificial ponds or container. Raceway ponds are the most commonly used artificial system (Jiménez et al., 2003). They are typical made of a closed loop, oval shaped recirculation channels, generally between 0.2 and 0.5 m deep, with mixing and circulation required to stabilized algae growth and productivity. Raceways pound are usually built in concrete, or compact earth and may be lined with white plastic. In a continuous production cycle, algae broth and nutrients are introduced in front of the paddlewheel and circulated through the loop to the harvest extraction point. The paddlewheel is continuous operation to prevent sedimentation. The microalgae's carbon dioxide requirement is usually satisfied from the surface air, but submerged aerators may be installed to enhance carbon dioxide absorption (Terry and Raymond, 1985).

Advantages and Disadvantages of Open Pound

Compared to closed photobioreactors, open pound is the cheaper methods of large-scale algal biomass production (Chisti, 2008). They also have lower energy input required (Rodolfi et al., 2009) and regular maintenance and cleaning are easier (Ugwu et al., 2008). Therefore, they may have the potential to return large net energy production (Rodolfi et al., 2009). However, the open pound system, require highly selective environments due to inherit threat of contamination and pollution from other algae species and protozoa (Pulz and Scheibenbogen, 1998). Monoculture cultivation is possible by maintenance of extreme culture environment, although only a small number of algae strains is suitable. Some example of extreme culture environment can be the species *Chlorella* adaptable to nutrient rich-media, *D. salina* adaptable to very high salinity and *Spirulina* adaptable to high alkalinity (Borowitzka, 1999). In respect with biomass productivity, open pound systems are less efficient when compared with closed photobioreactors (Chisti, 2007). This can be attributed to several determining factors, including, evaporation losses, temperature fluctuation in the growth media, diffusion of carbon dioxide in atmosphere, inefficient mixing and light limitation (Brennan and Owende, 2010; Ugwu et al., 2008). The evaporation losses make net contribution to cooling. It may also result in significant changes in ionic composition of the growth medium with interferences in algal growth (Pulz, 2001). The temperature fluctuation due to diurnal cycles and seasonal variations are difficult to control in open pound (Chisti, 2007). In addition, poor mixing by inefficient stirring mechanism, may result in poor mass carbon dioxide transfer rates causing low biomass productivity (Ugwu et al., 2008). Finally, the light limitation due to top layer thickness may also induce reduced biomass productivity. However, enhancing light supply is possible by reducing layer thickness: using thin layer inclined types of culture system, and improved mixing can minimise impacts to enhance biomass productivity (Brennan and Owende, 2010; Chisti, 2007; Pulz, 2001; Ugwu et al., 2008).

1.10.2 Closed Photobioreactor System

Closed systems include the horizontal tubular systems (Gudin and Chaumont, 1983; Molina Grima et al., 1994a), tubular reactors (Watanabe and Saiki, 1997), cascade reactors (Doucha and Lívanský, 1995), alveolar flat panels (Tredici, 2002), vertical flat panels (Samson and Leduy, 1985) and column photobioreactors (Mirón et al., 1999). However, in this part of the thesis will described only the horizontal tubular reactors and vertical flat panels, that were the two kind of closed photobioreactors tested. For information about the other closed photobioreactors, can be refer to the previous citation.

Horizontal Tubulars Photobioreactors

A tubular photobioreactor consists of an array of straight transparent tubes that are usually made of plastics or glass (Chisti, 2007). The tubular array, or solar collector, serves to sunlight capture and can be aligned horizontally (Molina Grima et al., 2003), vertically (Mirón et al., 1999), inclined (Ugwu et al., 2002) or as helix (Watanabe and Saiki, 1997). The tubes are generally 0.1 m or less in diameter (Chisti, 2007). Tube diameter is limited because light does not penetrate too deeply in the dense culture broth that is necessary for ensuring a high biomass productivity of the photobioreactors (Chisti, 2007). Algae cultures are re-circulated from a reservoir (i.e. the degassing column) to the solar collector and back to the reservoir, with a mechanical pump or air-lift pump (Chisti, 2007; Eriksen, 2008). Agitation and mixing are very important to prevent the biomass sedimentation in the tubes and to encourage gas exchange (Brennan and Owende, 2010; Chisti, 2007). In fact, photosynthesis generate oxygen and under high irradiance, the maximum rates of oxygen generation in a typical tubular photobioreactors may be $10 \text{ g m}^{-3} \text{ min}^{-1}$ (Chisti, 2007). To prevent inhibition and damage, the maximum tolerable dissolved oxygen level should not generally exceed about 400% of air value. Oxygen cannot be removed within a photobioreactors tube, so the culture must be periodically return to degassing zone, that is bubbled with air to strip out the accumulated oxygen. In addition, to remove the accumulated dissolved oxygen, the degassing zone must disengage all the gas bubbles from the broth so that essentially bubble-free broth returns to solar collectors tube (Chisti, 2007).

Vertical Flat Panels

Some of the earliest close systems are flat-plate photobioreactors (Samson and Leduy, 1985), which have received much research attention due to large surface area exposed to illumination (Ugwu et al., 2008) and high densities of photoautrophic cells observed ($>80 \text{ g L}^{-1}$) Hu et al. (2008). Recently, a new design of vertical flat panel photobioreacotor consisting of a plastic bag located between two iron frames has been proposed (Sierra et al., 2008; Tredici and Rodolfi, 2004); this brings a cost reducing to this type of photobioreactors. For the mixing and aeration of the culture, a gas sparger is usually placed inside, at the bottom of the plastic bags (Sierra et al., 2008).

Advantages and Disadvantages of Closed Photobioreactors

Microalgae production based on closed photobioreactor technology is designed to overcome some of the major problems associated with the described open pond production system. For example, pollution and contamination risks with open pond system, preclude their use in the pharmaceutical and cosmetic industry (Ugwu et al., 2008). The closed photobioreactors are more appropriate for sensitive strains as the closed configuration makes the control of potential contamination easier. However, the cost of closed systems are substantially higher than open pond system (Carvalho et al., 2006) and tubulars have design limitation on length of the tubes, which is dependent on potential oxygen accumulation, carbon dioxide depletion and pH variations in the systems. Therefore, they cannot be scaled indefinitely; hence, large scale productions plants are based on integration of multiple reactors units (Brennan and Owende, 2010; Eriksen, 2008). Besides, horizontal photobioreactors also have a better angle for incident light compared to vertical tubular reactors, allowing for more efficient light harvesting. This characteristic made tubular photobioreactors more suitable for outdoor mass cultures. However, this also causes the generation of large amount of heat, requiring sometimes expensive temperature control systems (Brennan and Owende, 2010; Richmond, 1987). The new design of vertical flat panels develop by Sierra et al. (2008); Tredici and Rodolfi (2004) may present a significant challenge at large scale operations. Nevertheless, they suffer for inadequate mixing and frequent "culture crashing" and are inherently fragile Wang et al. (2012). It was also observed that scale up using larger bags volumes do not necessarily lead to increased productivity (Martínez-Jerónimo and Espinosa-Chávez, 1994).

1.11 Objective of the Thesis

The objective of this thesis were correlate to a research project of the group of "Biotecnología de las microalgas marinas" of the University of Almería coordinate by Prof. Emilio Molina Grima. The aimed of this thesis was made a general study about the biomass production of oleaginous microalga *Nannochloropsis gaditana* CCMP527 aimed to biodiesel production, in outdoor and continuous mode. Therefore were study the effects of dilution rate combined with some technical features in three biophotoreactors types, on biomass volumetric productivity ($\text{g L}^{-1} \text{d}^{-1}$). In practice, the technical features were: (a)the air flow quantity in tubulars for study if there are some positive aspect in biomass productivity a scalar air flow in the degassing column;(b)the presence of sump in raceways, for evaluated if there are a positive aspects on biomass productivity due to eventually lower carbon dioxide losses;(c)and finally if the orientation and the shading effects in flat panels has some positive aspects on biomass productivity. Besides also a biochemical profile of biomass were performed in all photobioreactors with a particular attention on fatty acids profile, that is the most interesting matter for study biodiesel quality of microalgae, in three different photobioreactors.

Material and Methods

2.1 Organism and Culture Medium

The microalga *Nannochloropsis gaditana* Lubián CCMP527 obtained from the marine culture collection of the Institute of Marine Science of Andalucía-CSIC Cádiz (Spain), was used.

The experimental trial was carried out firstly in indoor condition and then in outdoor. The indoor was made on Algal medium (Bionova S.L., Santiago de Compostela, Spain) using 5 L glass bottle, at 20 °C with air injection of 0.2 vvm, under continuous illumination by fluorescent lamp (Hydro 920, Disano Illuminazione s.p.a., Rozzano, Italy) providing a light intensity of 1500 $\mu\text{mol m}^{-2} \text{s}^{-1}$ on the vessel surface. The component of Algal medium are reported by San Pedro et al. (2013) (Table 2.1).

Table 2.1: Component of modified Algal medium in indoor condition

Component	Concentration (g L^{-1})
NO_3^-	$4.9 \cdot 10^{-1}$
PO_4^{2-}	$4.0 \cdot 10^{-2}$
SO_4^{2-}	$9.0 \cdot 10^{-5}$
K^+	$3.1 \cdot 10^{-1}$
Ca^+	-
Mg^+	-
Fe^{2+}	$2.4 \cdot 10^{-3}$
Cu^{2+}	$6.0 \cdot 10^{-6}$
Co^{2+}	$4.5 \cdot 10^{-5}$
Zn^{2+}	$3.0 \cdot 10^{-4}$
Mn^{2+}	$3.7 \cdot 10^{-4}$
B^{3+}	-
Mo^{6+}	$5.4 \cdot 10^{-4}$
V^{5+}	-
EDTA	$9.4 \cdot 10^{-3}$

The outdoor growth was made putting 10 L of indoor culture and 90 L of Algal medium in 100 L glass column.

2.2 Outdoor Culture System

The experimental pilot plant, was located in the University of Almería, Almería (36° 48' N, 2° 58' W) (Spain). The pilot system was 330 m² of surface, necessary for a medium scale plant for the microalgal cultivation. Twelve photobioreactors were installed outside: three tubulars with a capacity of 340 L, three raceways of 850 L and six flat panels of 270 L. The technical features of tubulars and raceways were quite similar to a normal photobioreactors of these types, already described in the introduction section. The degassing column and the horizontal array of tube in tubulars were made in glass and a pump recirculate the liquid through the array of tube to the degassing column. Besides, the three tubular were equal.

In the plastic raceway reactors (AqualAgae S.L., Almería, Spain) were placed a motor linked to a mill, and a membrane air bubbler was placed before the mill. Two sumps with a capacity of 150 L and 75 L were installed in the two raceways and the air bubble was placed inside. A third raceways was without sump.

The basic layout of the flat panels used in the pilot plant has been described by Sierra et al. (1997). Briefly, a prolipoline bag was located between two iron frames at 0.070 m of distance. The two frames and plastic bag were 1.5 m high and 2.5 m long with a capacity of 250 L of volume. The plastic bag was made of free-disperdant 0.75 μ L polyethylene, with a transparency index of 0.95. An air diffuser in metal with 1 holes every 3 cm was placed side to side, at the bottom of the bags. The six flat panels were placed in two different orientation, three in direction North-South and the other three East-West.

Furthermore, to simulate the shadows effect, black panels were placed at three different distance between the single flat panel: 1.5 , 1.0 and 0.5 m. The black panels were placed in East-West and North-South oriented flat panels.

In all of the twelve photobioreactors were placed probes (Crison Instruments, Spain) for the measure of pH and temperature, and in the tubulars and raceways were placed also probes for the measure of dissolved oxygen percentage (% DO). The data of air flow incoming and outcoming all the probes were captured by a control panel installed for each photobioreactors and by a PC computer for on-line registration. Finally, a ball vulvae was located overhead in all of twelve photobioreacotrs allowing the harvesting during continuous mode operation.

2.3 Operating Mode

The photobioreactors were inoculated with outdoor culture. In tubulars and flat panels were placed 50 L of outdoor culture whereas in raceways 100 L. The remaining volume were filled with culture medium. When the the steady state was reached, the dilution was started. Each photobioreactor was operated as a chemostat for six months as continuous mode at different dilution rate: 0.33 d⁻¹, 0.28 d⁻¹ and 0.12 d⁻¹. Fresh medium was added at a constant rate dilution during the daylight and every

morning were performed analysis to determinate the biomass concentration, the pH and the status of the cells with F_v/F_m ratio. The percentage of oxygen and temperature data were took by PC-Computer, that calculated the medium value of the day.

A modified Algal medium (San Pedro et al., 2013), were prepared with an automatic system of fertirrigation (Nutritec 9000, Riegos y Tecnología S.L., Águilas, Spain) that mixed the sea water with the macro and micro nutrients. The components of the medium used are report in Table 1.

The medium was sterilized with ozone by an ozone generator (Riegos y Tecnología S.L., Águilas, Spain) and passed through three filters of 10 μm , 5 μm and 1 μm (1N2FP, 3M Filter Housing, Cergy-Pontoise, France) to remove particles and cell of this size.

The cultures were maintained around pH 8 by injection on demand of carbon dioxide. The air flow rate entering in photobioreactors was regulate using suitable valves and flowmeter, placed inside the control panels. The air flow in raceways was maintained at 20 L min^{-1} and in flat panels 41 L min^{-1} . Instead, the air flow in the three tubulars were different: 20 L min^{-1} in tubular one, 26 L min^{-1} in tubular two and 32 L min^{-1} in tubular three.

2.4 Analytical Methods

2.4.1 Volumetric Productivity

Every morning analysis to determinate the volumetric were performed. The analysis were performed until the volumetric productivity was almost constant for four consecutive day (i.e. the steady state was reached). The volumetric productivity of the biomass (P_b) was calculated using the following equation:

$$P_b \text{ (g L}^{-1} \text{ d}^{-1}) = C_b \times D$$

where C_b is the biomass concentration and D is the used dilution rate.

The Biomass concentration (C_b) was evaluated measuring the optical density (OD) at 750 nm using a 1 cm cuvette (DR/4000U Spectrophotometer; Hach Company, Loveland, CO, USA) and was established :

$$C_b \text{ (g L}^{-1}) = m \times OD_{750} \times FD$$

where m is the slope of the calibration line C_b vs OD_{750} , obtained from dry weight values and their dilution, OD_{750} is the optical density at 750 nm and FD is the dilution factor used for the measure.

2.4.2 Determination F_v/F_m

The F_v/F_m ratio was determinate using a fluorometer (AquaPen-C AP-C 100; Photon Systems Instruments, Brno, Czech Republic). Samples were placed in 4 mL cuvettes, in dark conditions for 3 minutes in order to measure the potential PSII photochemical efficiency.

2.4.3 Dry Weight

To determine dry weight of biomass, 600-1600 mL samples were taken from each of the twelve photobioreactors, to obtain at least 400 mg of biomass for all biochemical analysis. Firstly, samples were centrifuged at 7500 rpm for 6 min (Sigma 4-15 4K15; Sartorius Group, Goettingen, Germany) to remove sea water. Then biomass was washed twice with distilled water to eliminate remained sodium chloride. Subsequently, biomass samples were placed in a previously weighed Petri dish and freeze-dried (LyoQuest; Telstar, Terrassa, Spain). Dry weight was evaluated as difference between the weight of Petri dish with freeze-dried biomass and empty Petri dish.

The biochemical composition of biomass was analysed following the methods described below. Previously to each analysis the biomass was grinded in a mortar for 5 min with aluminium oxide in a 1:1 ratio.

2.4.4 Total Lipids

The total lipid content in biomass was determined gravimetrically following Kochert (1978) method. According to this, 2 mL of chloroform:methanol (2:1, v/v) mixture was added to 100 mg of biomass previously placed in a tube, which was shaken and centrifuged (Digicen 21R; Ortoalresa, Madrid, Spain) for 5 min at 3500 rpm. The supernatant was collected and the biomass subjected to a second extraction with 1 mL portion of chloroform-methanol mixture. This procedure was repeated until supernatant occurred clear. To remove proteins from supernatant collected 3 mL of 0.1 N hydrochloric acid and 0.3 mL of 0.5% magnesium chloride were added to each tube. To achieve phase separation solution was centrifuged again. Bottom phase containing lipids, was transported to previously weighed tube. Solvent was evaporated under nitrogen stream at 45 °C.

Percentage content of lipids in biomass was calculated as follows :

$$\text{Lipid (\% w/w)} = \frac{W_{\text{lipid}}}{m} \times 100$$

where W_{lipid} is the weight of the total lipid in mg (i.e. the difference between dry tube and empty tube), m is the biomass weighed initially in mg.

2.4.5 Proteins

Proteins content in biomass was evaluated by a modification of the Lowry et al. (1951) as described by González López et al. (2010).

To disrupt cell wall and release proteins, 9 mL of Lysis buffer reactive were added to 20 mg of biomass. The solution was kept at 25 °C for 30 min. Afterwards, 100 μ L of sample were taken into an eppendorf tube with 900 μ L of Lysis buffer. Samples were shaken and diluted taking 100 μ L of each samples into another eppendorf tube with sodium dodecyl sulfate (SDS) reactive. The Folin and Lowry reagents were prepared. The Folin reagent was prepared diluting the Folin and Ciocalteu's phenol reagent in distilled water (Folin:water ratio 1:1). The Lowry reagent was prepared through three reagents : Reagent A (4.0 g L⁻¹ of sodium hydroxide and 20 g L⁻¹

of sodium carbonate), Reagent B1 (0.001 g L⁻¹ of copper II sulfate pentahydrate) and Reagent B2 (0.002 g L⁻¹ of potassium sodium tartrate tetrahydrate) in ration 100:1:1.

At this point, 1 mL of Lowry reagent was added in each eppendorf and immediately shaken. After 10 minutes of Lowry reagent insertion, 100 µL of Folin reagent were added and shaken immediatly.

The samples were kept under dark for 30 minutes. Then, the assorbance was measured with a spectrophotometer analysis (Helios Omega UV-Vis, Thermo Fisher Scientific Inc.; Waltham, MA, USA) at a wavelenght of 750 nm.

In order to determinate the protein concentration (C), a calibration curve was done using a bovine serum albumin (BSA) as protein source (Table 2).

Table 2.2: BSA concentration used for the calibration curve of proteins

Sample	BSA (mL)	Lysisbuffer (mL)	[BSA] (µg mL ⁻¹)
White	0	5	0
1	0.0625	4.9375	25
2	0.125	4.875	50
3	0.25	4.75	100
4	0.3125	4.6875	125
5	0.625	4.375	250
6	1.25	3.75	500
7	1.88	3.12	750
8	2.5	2.5	1000

Proteins content was calculated as follows :

$$\text{Protein (\% w/w)} = \frac{C \cdot V \cdot D}{m} \times 100$$

where C is the concentration of protein (mg L⁻¹), acquired from the calibration curve, V is the volume (L) of lysis buffer used for diluting the samples, D is the dilution factor and m (mg) is the amount of biomass.

2.4.6 Fatty Acids

A modified protocol of Rodríguez-Ruiz et al. (1998) was used for determine fatty acids. Biomass was placed in a pirex tube and fatty acids were extracted using hexane as solvent, nonadecanoic acid (C19:0) as pattern, and acetyl chloride-methanol 5:100 (v/v) as metilant agent.

Tubes were shaken and placed in a thermoblock (JP Selecta S.A., Abrera, Spain) at 105 °C for 10 minutes. After this time, the tubes were shaken again and kept in the thermoblock for another 10 minutes.

Once the tubes were at air temperature, 1 mL of distilled water was added. Then the tubes were centrifuged (Digicen 21R; Ortoalresa, Madrid, Spain) at 3500 rpm for 5 minutes.

The hexane phase containing the fatty acids, was taken and placed in tubes for gas chromatography analysis. The identification of the different peaks reported by the chromatogram was carried out through a fatty acids pattern, which contained every fatty acids we needed to screen. In order to quantify the fatty acid content, the peaks were compared with the nonadecanoic acid one, as we knew its concentration.

The content of any fatty acid i in biomass was calculated as follows:

$$\text{Fatty Acid}_i (\% \text{ w/w}) = \frac{0.125 \cdot A}{m} \times 100$$

where 0.125 is the nonadecanoic acid concentration in the tube, expressed in mg, A is the ratio between the area of the peak for the fatty acid i and the nonadecanoic acid peak area, m is the amount of biomass (mg).

The total quantity of fatty acids in biomass was calculated as follows:

$$\text{Fatty Acid}_{\text{total}} (\% \text{ w/w}) = \sum_{k=1}^n \text{Fatty Acid}_i$$

where Fatty Acid_i is the content of any fatty acid i in biomass. While, the amount of any fatty acid i respect to the total fatty acid was calculated as follows:

$$\% \text{ single Fatty Acid}_i = \frac{\text{FA}_i}{\text{FA}_{\text{total}}} \times 100$$

where FA_i is the content of any fatty acid i in biomass, FA_{total} is the total content of fatty acids in biomass.

2.4.7 Ashes

Ashes were determined by calcination. Biomass was placed in previously weighed crucible and incubated in oven at 500 °C for 24h. Then, samples were kept in desiccator until they reached the air. Ash content was determined gravimetrically as follows:

$$\text{Ash} (\% \text{ w/w}) = \frac{W_{\text{ash}}}{m} \times 100$$

where W_{ash} is the weight of the ash in mg (i.e. the difference between full crucible and empty crucible), m is the biomass weighed initially in mg. For obtain the ash free dry weight (AFDW) for each biochemical compound in biomass, was calculated as follows:

$$\text{AFDW}_i (\% \text{ w/w}) = \% C_i \times \frac{100}{100 - \% \text{ Ash}}$$

where AFDW_i is the content of any biochemical compound i ash free in biomass, C_i is the content of any biochemical compound i in biomass, Ash is the content of ash.

2.5 Solar Irradiance Measurement

2.5.1 Global Solar Irradiance

A solar irradiator sensor/pyranometer (LP02, Hukseflux Thermal Sensor B.V., Delft, The Netherlands) was placed in the pilot system to determine the global radiation at ground level. The data were process by a PC computer to obtain the daily mean value of global irradiation.

2.5.2 Average Irradiance

The average irradiance (I_{av}) was measured periodically using a quantum scalar irradiance meter (QSL-100 Biospherical Instruments Inc., San Diego, CA, USA). In flat panels the measure were made in the upper-part, medium-part and lower-part of the bags. The medium value was the avarage irradiance of flat panel.

2.6 Statistical Analysis

A two-ways ANOVA test were performed for analyse the data, and a probability P value of less then 0.05 was considered significant. When the F ratio of the interaction was significant difference (LSD) was utilised to compare different mean for $P = 0.05$.

Result

3.1 Environmental data

The optimal growth of microorganism is conditioned by the environmental factors like pH, temperature and light intensity. This could be easier in indoor conditions, where factors like temperature and light can be regulated, but in outdoor conditions, light and temperature depend on the latitude and the season, so they are not manageable factors. The experiments were performed during the winter season but the average global irradiation in each dilution rate experiments, were high due to latitude of Almería (36° 48' N, 2° 54' W) near the Tropic of Cancer and in semi-arid climate. Global irradiation ranged among 1087 (± 253.2), 1332 (± 476.2) and 1452 (± 150.8) $\mu\text{mol m}^{-2} \text{s}^{-1}$ at dilution rate of 0.33 d⁻¹, 0.28 d⁻¹ and 0.12 d⁻¹ respectively. Also temperatures inside the photobioreactors were not so low for algae growth. Nevertheless, like reported by Rocha et al. (2003), the temperature was optimal for the *Nannochloropsis* growth's (25 \pm 5 °C) only in Tubular photobioreactors with a medium daily value during the three dilution rate experiments of 21 \pm 2.5 °C. While, in raceways and flat panels East-West and North-South the temperature were 14 \pm 2 °C, 16 \pm 1.9 °C and 16 \pm 2.5 °C respectively. This data advising that the tubulars photobioreactors are more suitable for outdoor mass cultures since expose a larger surface area to sunlight. Microalgae are sensitive to the pH change, so this control is essential for keeping high growth rates. The pH was maintained at optimal level, around 8 (Rocha et al., 2003), in every biophotoreactors with the injection on demand of carbon dioxide.

3.1.1 F_v/F_m Ratio

The F_v/F_m ratio is an important parameter that represents the maximal quantum yield for PSII photochemistry (Björkman and Demmig, 1987). Like reported to McMinn and Hegseth (2004) a value around 0.6 it can be considered a good level of efficiency of photosynthetic apparatus (F_v/F_m o microalga), while under the threshold of 0.5 it can be considered the culture photoinhibit. The measure of F_v/F_m was made every morning putting the microalgae in dark for 3 min, before the reading with the fluorometer. As reported in figures 1 to 4 in all type of photobioreactors at

different dilution rates, the F_v/F_m ratio were above the threshold value of 0.5 indicating that not photoinhibition process occurred in these experimental conditions (Figure 3.1).

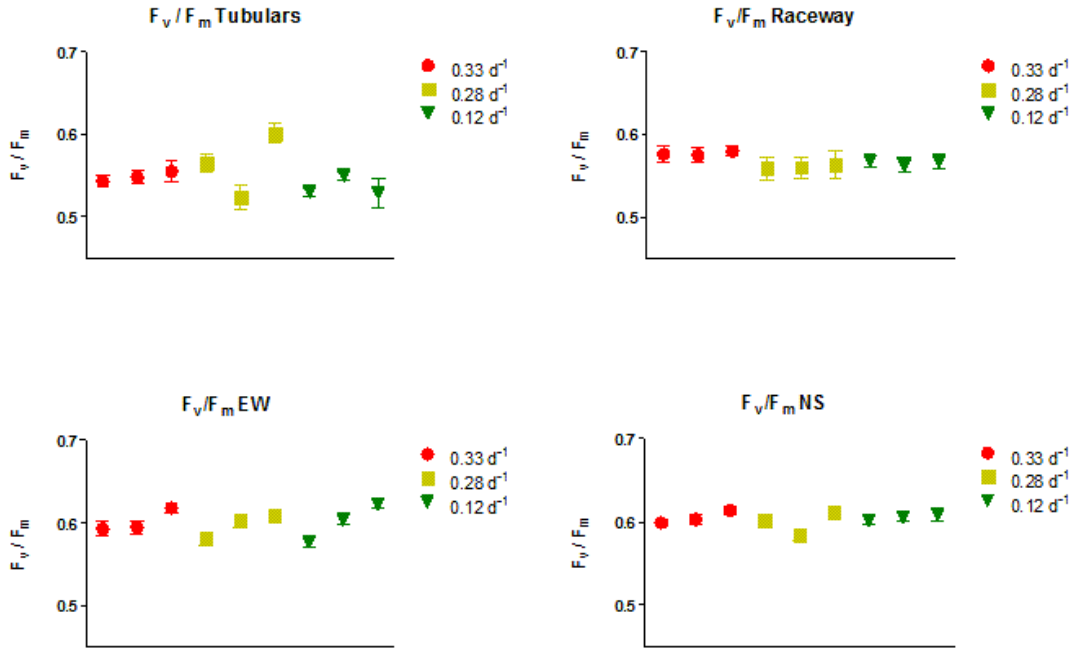


Figure 3.1: Medium value of F_v/F_m in photobioreactors at dilution rate of 0.33 d^{-1} , 0.28 d^{-1} and 0.12 d^{-1} . Tubular: for each dilution the first point indicates an air flow of 20 $L\ min^{-1}$, the second 26 $L\ min^{-1}$ and the third one 32 $L\ min^{-1}$. Raceway: for each dilution the first point indicates an sump volume of 150 L, the second 75 L and 0 L. Flat Panels: for each dilution the first point indicates a distance from black panel of 1.5 m, the second 1 m and the third one 0.5 m.

3.2 Biomass production

Effect of dilution rate in different type of photobioreactors (Figure 3.2) was investigated in outdoor and continuous mode.

The initial inoculum was placed indoor with halogens lamp and aerated with air 0.2 vvm, and later in outdoor column. The photobioreactors were inoculated with different quantity of culture and the dilution started when the steady state was reached.

In every photobioreactors the dilution 0.28 d^{-1} (i.e. specific growth rate $\mu = 0.28\ d^{-1}$) was significant for the biomass volumetric production P_b ($g\ L^{-1}\ d^{-1}$). The tubulars were the photobioreactor that reached the highest value of biomass productivity $0.52 \pm 0.076\ g\ L^{-1}\ d^{-1}$.

From obtained results it is evident as different parameters influenced the biomass productivity, like air flow in tubulars, presence or not of sump and the dimension of this latter in raceways, orientation and distance between them in flat panels. For

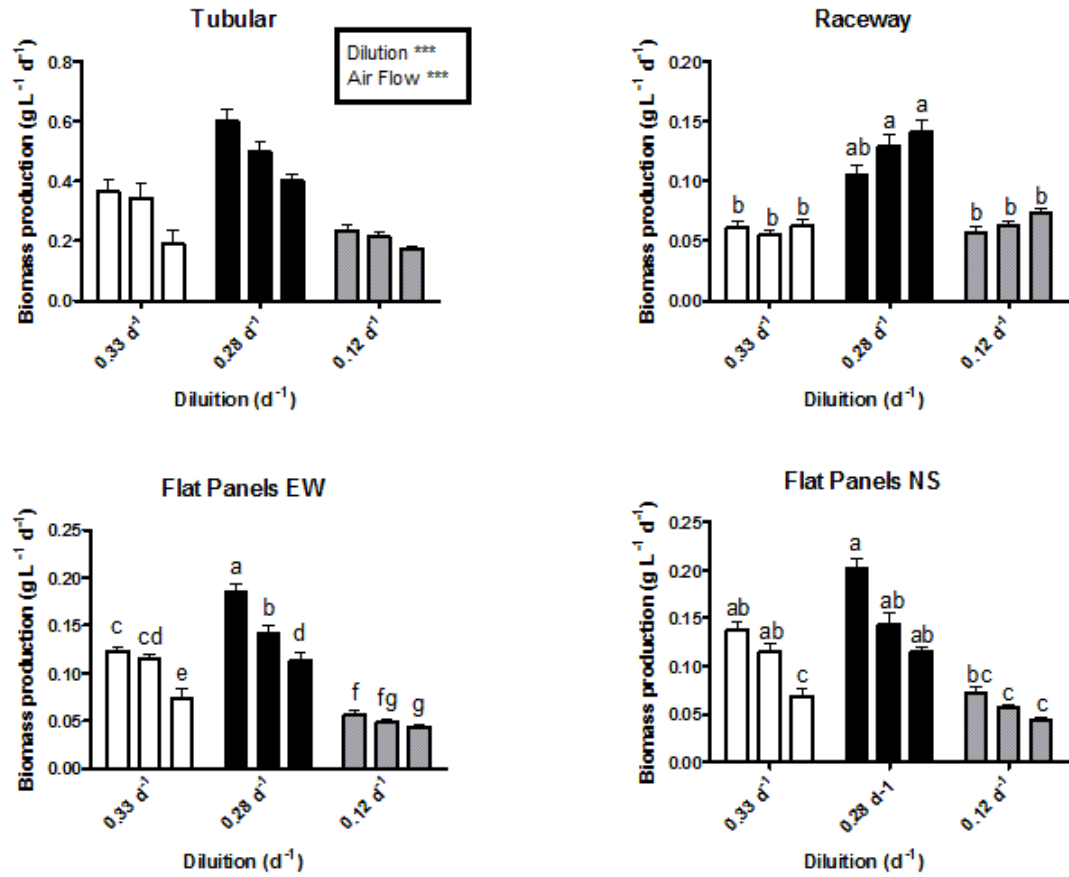


Figure 3.2: Biomass production of *Nannochloropsis gaditana* in different photobioreactor at different dilution. Tubular: for each dilution the first bar indicates an air flow of 20 L min $^{-1}$, the second 26 L min $^{-1}$ and the third one 32 L min $^{-1}$. Raceway: for each dilution the first bar indicates an sump volume of 150 L, the second 75 L and 0 L. Flat Panels: for each dilution the first bar indicates a distance from black panel of 1.5 m, the second 1 m and the third one 0.5 m. Each value represents the mean of 4 replicates (\pm standard deviation). For Tubular graph the absence of letters indicate no significant interactions between the two factors. In this graph the significance of single factor is reported. Means followed by the same letters are significantly different ($P=0.05$) in accordance with two-ways ANOVA test

this reasons, following we reported the effects of environmental factors in different types of photobioreactors.

Tubulars

On the basis of two-way ANOVA test, the interaction between dilution rate and air quantity in tubulars was not significant for biomass productivity. In fact, the trend of biomass production at different dilution rate were quite similar. Meanwhile, a significant difference between the single bars was observed. So the biomass productivity was influenced singularly, by the dilution rate and air flow. The oxygen dissolved in closed photobioreactors like tubulars, is the major problem: high level of oxygen could produce photooxidative damage and inhibit the carbon dioxide utilization by the Rubisco enzyme (Wang et al., 2012). Three different air flow was utilized in tubulars to optimizing the degassing efficiently remove the oxygen in the culture broth (Table 3.1).

Table 3.1: Percentage of oxygen dissolved in tubulars at dilution rate of 0.33 d^{-1} , 0.28 d^{-1} and 0.12 d^{-1} . Means followed by the same letters are not significantly different of $P = 0.05$

Air Flow	% Oxygen Dissolved $\pm \sigma^2$		
	20 L min^{-1}	26 L min^{-1}	32 L min^{-1}
0.33 d^{-1}	239 ± 16	185 ± 23.4	191 ± 14
0.28 d^{-1}	256 ± 29	226 ± 38.2	241 ± 41.1
0.12 d^{-1}	245 ± 23.3	220 ± 21.8	230 ± 12.8
\bar{X}	247 a	210 b	221 b

The rate of oxygen dissolved was around 200%, so in all tubulars the air flow injected was optimal for the degassing processes and lower level was observed with air flows of 26 L min^{-1} and 32 L min^{-1} than 20 L min^{-1} .

Raceways

A relation between the sump volume and the dilution rate for biomass productivity was found in raceways photobioreactors (Figure 3.1). The high level of production was in raceway with no sump and 70 L of sump at dilution rate of 0.28 d^{-1} , where the biomass volumetric productivity was quite similar.

Flat Panels

In flat panels the effects of orientation and shading with neighbouring flat panel, on biomass production was studied. The average irradiance (I_{av}), that is a estimation of irradiance inside the culture (Molina Grima et al., 1996), was established with the biospherical instrument.

As already reported by (Sierra et al., 2008) the irradiance that is intercepted by flat panels was influenced by the latitude. In fact, the East-West orientated

flat panels intercepted more solar radiation than the North-South for latitude above 35°N to the equator, while for latitude below 35 °N was better the North-South orientation. However, considering the latitude of Almería (36°N), no significant difference of average irradiance was observed between the flat panels East-West and North-South (Table 3.2).

However a decisive influence on biomass production was attributable to the distance from the black panels, that was indicative of intercepted irradiance by flat panels (Table 3.2; Figure 3.2). The high biomass production in flat panels East-West and North-South was at distance from black panels of 1.5 m, at dilution rate of 0.28 d⁻¹.

Table 3.2: Average irradiance of Flat Panels East-West and North-South at different distance of 1.5, 1.0, and 0.5 m from the black panels. Means followed by the same letters are not significantly different of P = 0.05

Distance (m)	$\mu\text{mol m}^{-2} \text{s}^{-1} \pm \sigma^2$		
	1.5	1	0.5
EW	595 \pm 102.1	437 \pm 67.5	259 \pm 64.1
NS	660 \pm 197.8	416 \pm 87.4	289 \pm 90.3
\bar{X}	632 a	426 b	274 c

3.3 Biochemical of Biomass

The biochemical composition of biomass were performed using the analytical methods described before, and for each value the amount of ash was subtracted. The total lipids ash free content in biomass at different dilution rate was quite similar, around 29 - 20 % in all photobioreactors (Figure 3.3). Also the proteins content at the three dilution rate (Figure 3.4) were around 40 - 48 % of total ash free biomass, in all photobioreactors. The fatty acids ash free contents (Figure 3.5) were higher in tubulars than other photobioreactors at dilution rate of 0.12 d⁻¹, with a percentage that reached 21 - 23 % of total lipids contents. The fatty acid profile at steady state, showed a higher amount of saturate fatty acids (SAFs) than polyunsaturated (PUFAs) in almost all photobioreactors. However, the high content of eicosapentaenoic acid (C20:5n3) in PUFAs, confirm that this microalgae is a good producer of this fatty acid (Table 3.3 to Table 3.6).

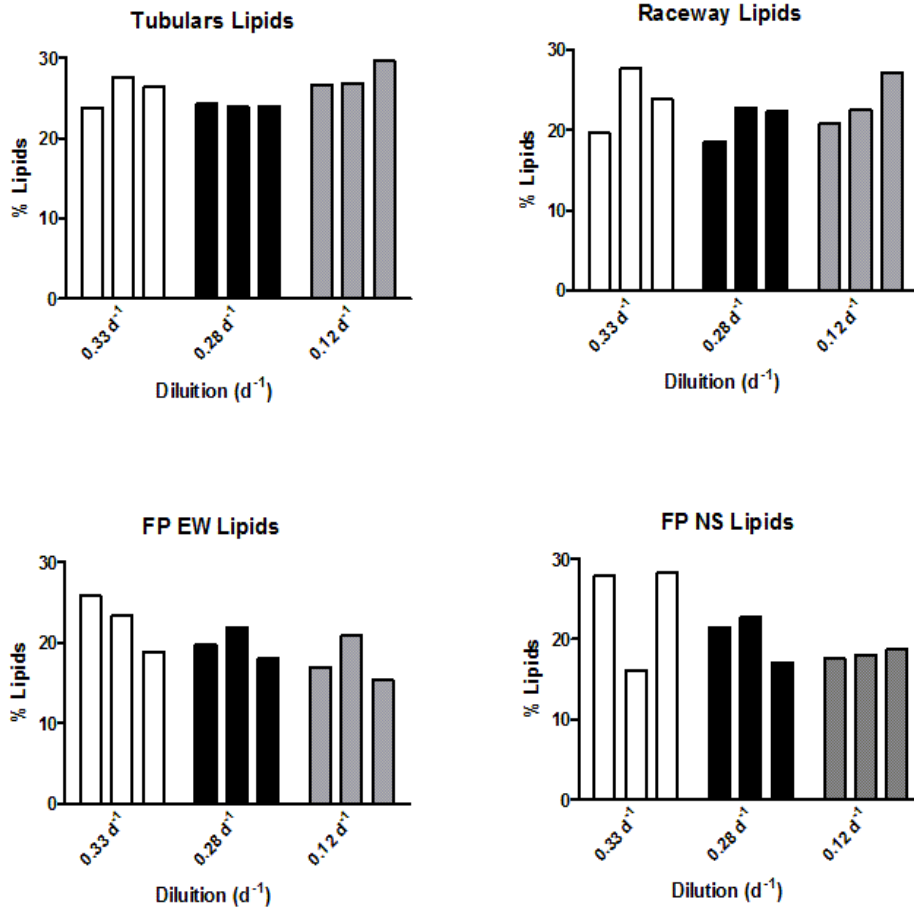


Figure 3.3: Percentage of lipids in different photobioreactors at dilution rate of 0.33 d⁻¹, 0.28 d⁻¹ and 0.12 d⁻¹. Tubular: for each dilution the first bar indicates an air flow of 20 L min⁻¹, the second 26 L min⁻¹ and the third one 32 L min⁻¹. Raceway: for each dilution the first bar indicates an sump volume of 150 L, the second 75 L and 0 L. Flat Panels: for each dilution the first bar indicates a distance from black panel of 1.5 m, the second 1 m and the third one 0.5 m.

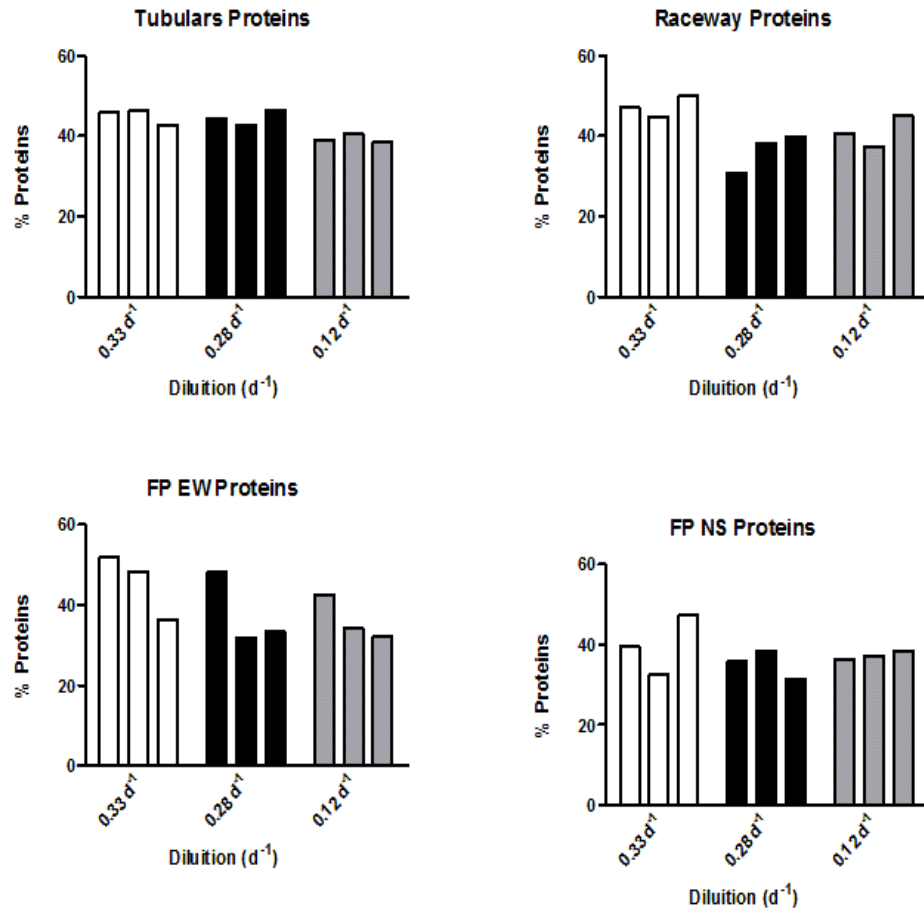


Figure 3.4: Percentage of proteins in different photobioreactors at dilution rate of 0.33 d⁻¹, 0.28 d⁻¹ and 0.12 d⁻¹. Tubular: for each dilution the first bar indicates an air flow of 20 L min⁻¹, the second 26 L min⁻¹ and the third one 32 L min⁻¹. Raceway: for each dilution the first bar indicates an sump volume of 150 L, the second 75 L and 0 L. Flat Panels: for each dilution the first bar indicates a distance from black panel of 1.5 m, the second 1 m and the third one 0.5 m.

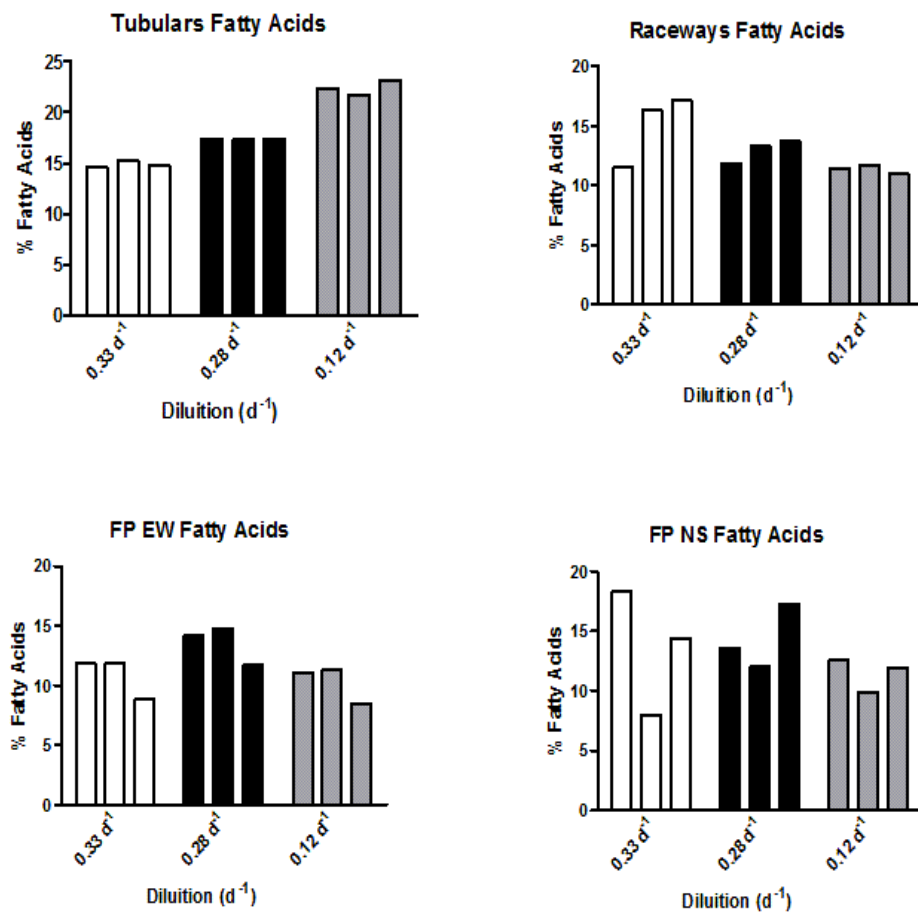


Figure 3.5: Percentage of fatty acids in different photobioreactors at dilution rate of $0.33 d^{-1}$, $0.28 d^{-1}$ and $0.12 d^{-1}$. Tubular: for each dilution the first bar indicates an air flow of $20 L min^{-1}$, the second $26 L min^{-1}$ and the third one $32 L min^{-1}$. Raceway: for each dilution the first bar indicates a sump volume of 150 L, the second 75 L and 0 L. Flat Panels: for each dilution the first bar indicates a distance from black panel of 1.5 m, the second 1 m and the third one 0.5 m.

Table 3.3: Fatty acid profile of tubulars. In this table were not reported the percentage of the standard (nonadecanoic acid).

		% Fatty Acid									
Fatty Acid		14:00	16:00	16:1n7	16:4n1	18:1n9	18:2n6	18:3n3	20:4n6	20:5n3	Others
0.20 L min ⁻¹	0.33 d ⁻¹	5.59	19.79	29.01	-	2.5	2.68	-	6.32	23.55	10.57
	0.28 d ⁻¹	4.64	22.70	24.18	3.58	5.95	2.39	3.96	4.51	15.65	12.44
	0.12 d ⁻¹	5.77	28.31	30.37	-	8.64	2.20	-	4.75	12.72	7.25
0.26 L min ⁻¹	0.33 d ⁻¹	6.29	22.48	32.72	-	3.76	2.64	-	4.39	19.65	8.06
	0.28 d ⁻¹	5.16	21.05	27.89	4.60	3.19	3.19	-	5.19	23.01	9.91
	0.12 d ⁻¹	6.28	28.35	28.60	-	10.13	2.34	-	3.97	14.37	5.96
0.32 L min ⁻¹	0.33 d ⁻¹	6.30	22.09	31.61	-	3.47	2.88	-	4.42	21.31	7.92
	0.28 d ⁻¹	4.83	23.91	25.51	3.50	5.54	2.18	3.80	3.57	15.71	11.46
	0.12 d ⁻¹	6.56	29.24	31.66	-	7.70	1.84	-	3.98	13.07	5.94

Table 3.4: Fatty acids profile of raceways. In this table were not reported the percentage of the standard (nonadecanoic acid).

Fatty Acid	% Fatty Acid										
	14:00	16:00	16:1n7	16:4n1	18:1n9	18:2n6	18:3n3	20:4n6	20:5n3	Others	
150 L	0.33 d ⁻¹	4.91	16.51	28.59	-	1.90	1.92	4.50	30.90	11.12	
	0.28 d ⁻¹	4.35	16.06	25.19	-	3.31	2.86	4.92	31.84	11.45	
	0.12 d ⁻¹	4.47	14.02	23.87	-	4.68	3.19	5.38	31.68	12.71	
75 L	0.33 d ⁻¹	5.38	19.37	33.36	-	2.32	2.06	3.80	24.51	9.21	
	0.28 d ⁻¹	4.21	15.90	24.87	-	2.97	2.90	5.06	33.36	10.37	
	0.12 d ⁻¹	4.37	13.45	23	-	4.71	3.46	5.60	33.88	11.52	
0 L	0.33 d ⁻¹	5.90	21.53	34.79	-	2.92	2.05	3.32	21.42	8.07	
	0.28 d ⁻¹	4.44	17.60	26.50	-	3.27	2.90	4.60	30.73	9.96	
	0.12 d ⁻¹	4.42	14.13	23.98	-	4.05	3.17	5.58	32.62	12.05	

Table 3.5: Fatty Acid Profile of Flat Panels East-West. In this table were not reported the percentage of the standard (nonadecanoic acid).

Fatty Acid	% Fatty Acid										
	14:00	16:00	16:1n7	16:4n1	18:1n9	18:2n6	18:3n3	20:4n6	20:5n3	Others	
1.5 m	0.33 d ⁻¹	4.23	14.91	25.46	-	2.05	1.96	5.83	34.73	10.83	
	0.28 d ⁻¹	3.64	15	20.50	-	4.50	3.44	5.44	35.40	12.09	
	0.12 d ⁻¹	5.29	13.41	20.19	-	4.15	4.25	5.66	37.72	9.33	
1.0 m	0.33 d ⁻¹	4.04	15.22	26.25	-	2.28	1.93	5.67	33.59	11.02	
	0.28 d ⁻¹	4.33	15.05	3.87	-	3.19	5.44	5.44	36.74	10.85	
	0.12 d ⁻¹	4.68	12.44	18.79	-	3.89	3.65	5.63	37.75	13.17	
0.5 m	0.33 d ⁻¹	4.07	14.77	25.68	-	2.12	1.82	5.50	34.88	11.16	
	0.28 d ⁻¹	4.39	15.12	20.99	-	3.51	2.78	5.42	36.29	11.50	
	0.12 d ⁻¹	4.59	12.35	19.07	-	3.69	3.31	5.49	38.70	12.80	

Table 3.6: Fatty Acid Profile of Flat Panels North-South. In this table were not reported the percentage of the standard (nonadecanoic acid).

		% Fatty Acid									
Fatty Acid		14:00	16:00	16:1n7	16:4n1	18:1n9	18:2n6	18:3n3	20:4n6	20:5n3	Others
1.5 m	0.33 d ⁻¹	8.69	19.28	32.55	-	2.01	1.98	-	6.05	19.89	9.56
	0.28 d ⁻¹	4.02	14.47	21.79	-	4.10	3.52	-	6.17	35.11	10.81
	0.12 d ⁻¹	4.52	14.85	23.11	-	4.69	3.78	-	6.78	32.68	9.64
1.0 m	0.33 d ⁻¹	4.19	14.62	24.89	-	2.25	2.07	-	5.58	34.81	11.60
	0.28 d ⁻¹	4.83	17.26	24.01	-	4.18	3.11	-	5.03	31.58	10
	0.12 d ⁻¹	5.42	14.99	20.74	-	5.16	4.61	-	5.64	35.64	7.80
0.5 m	0.33 d ⁻¹	4.46	17.21	29.35	-	2.93	2.10	-	5.83	27.42	10.69
	0.28 d ⁻¹	3.94	18.77	27.11	-	3.26	2.68	-	6.76	28.05	9.42
	0.12 d ⁻¹	3.72	15.11	24	-	3.34	2.89	-	7.22	31.42	12.29

3.4 Ash

The analysis of ash was made to evaluate the quantity of inorganic components of cells in the biomass (Borowitzka and Moheimani, 2012). This value were used for discard the inorganic matters in the biochemical components of our biomass.

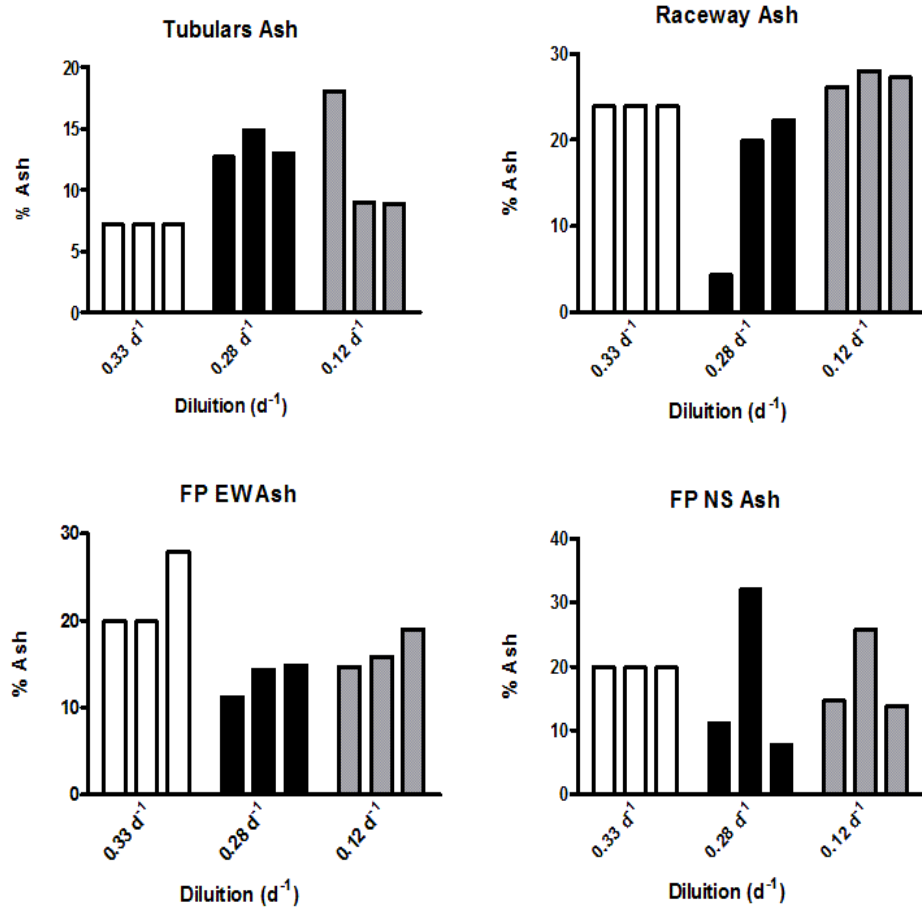


Figure 3.6: Percentage of ash in different photobioreactors at dilution rate of 0.33 d⁻¹, 0.28 d⁻¹ and 0.12 d⁻¹. Tubular: for each dilution the first bar indicates an air flow of 20 L min⁻¹, the second 26 L min⁻¹ and the third one 32 L min⁻¹. Raceway: for each dilution the first bar indicates an sump volume of 150 L, the second 75 L and 0 L. Flat Panels: for each dilution the first bar indicates a distance from black panel of 1.5 m, the second 1 m and the third one 0.5 m.

Discussion

4.1 Biomass Concentration

In this work a study about the biomass productivity by the microalga *Nannochloropsis gaditana* CCMP527 in three different type of photobioreactors (tubulars, raceways and flat panels) in outdoor and continuous mode at three different dilution rate was performed. As expected, tubulars were the photobioreactors that reached the higher biomass volumetric productivity than the others. This confirms the tubular are the best photobioreactors to produce algae biomass. Among different dilution rates utilized in all photobioreactors, that of 0.28 d^{-1} induced higher biomass productivity has compared to the others.

4.1.1 Tubular

In tubular photobioreactors no interaction between the dilution rate and air flow quantity on biomass production was observed but on other hand, a interaction between the single factors in biomass productivity was observed. The air flow selected for spit out the oxygen in tubulars was optimal in any case (Table 3.1). The interesting features was any dilution rate experiments , the volumetric productivity decrease to increased of the air quantity. This is could be by the effects of hydrodynamic stress in the microalgae cells. Gudín and Chaumont (1983) reported that the pumping of the culture can be involved in damage on microalgal cells. Like reported by Camacho et al. (2000) this can be happen also with the agitation of the culture by the air flow in the photobioreactors.

4.1.2 Raceways

In raceways we assayed if the carbon dioxide losses in a open pound systems can influenced the biomass production. In this case statistical analysis shown the positive interaction between the presence of sump or not and dilution rate in biomass productivity. However, no significant difference in biomass production were observed in raceways without sump of 150 L. Anyway, considering that the sump presence improved the volumes of the raceways, and this have a important effect on dilution

rate of the culture, the biomass production was high in raceways without sump. This results suggest that more study have be performed to evaluating the presence of the sump could be positive for the carbon dioxide losses. An important ratio of carbon dioxide inside the cell is the primary requirement to Rubisco enzyme for start the Calvin cycle, and so have a good biomass production.

4.1.3 Vertical Flat Panels

Even in flat panels the two-ways ANOVA test shown a significant interation between the distance from black panels and dilution rate with the biomass productivity. Besides, the data of average irradiance (Table 3.2) confirm that flat panels at a distance of 1.5 m received more light and consequently produce more biomass. These data could confirm the study of Zhang et al. (1999) in which they found that the reduction in distance, i.e. increasing the number of plates in a fix ground area, was not a efficient way to increased total productivity. However, the results obtained by Carlozzi (2003) was in complete disagreement. This can be explained considering the diluting solar radiation. Even Richmond and Zou (1999) proved the positive effects of an efficient utilization of solar irradiance. Culturing photoautotrophic microorganism outdoors is based, conceptually, on effective utilization of the high energy solar irradiance, which represents excessive radiation for the photosynthetic machinery of the individual cell. In fact, distribute sunlight at throughout the culture at intensity below the saturation point of the organism, it has the benefits to increase the photosynthetic efficiency and high is the growth rate and the volume output rate ($\text{g L}^{-1} \text{d}^{-1}$).

4.2 Profile of Biochemical Data

Lipid content and fatty acids composition are also subject to variability during the growth cycle (Hu et al., 2008). Increases in the relative proportion of both saturated (C16:0, C18:0) and mono-unsaturated fatty acids and decreased portion of polyunsaturates fatty acids in total lipids were also associated with grown phase transition from the logarithmic to steady state. However in some microalgae classes, like the *Eustigmatophyceae*, the eicosanpentanoic acid (C20:5n3) is the exception. In fact, during the steady state this algae genus stored high level of this polyunsaturated fatty acids. In this study the research confirmed this trend (Table 4.1), the saturated and monounsaturated fatty acids percentage was higher than the polyunsaturated fatty acids. Considering the engine performance described in the introductory section (Cetane number, heat of combustion, cold flow properties, oxidatively stability, viscosity, lubricity and exhaust emission) this data confirmed that fatty acid profile can be interesting for the biodiesel production. However, in raceways and flat panels the high level of the polyunsaturated acid in total fatty acids, could result in a no optimal growth. In fact temperature of the culture was under the ideal range reported by Rocha et al. (2003). Temperature has been found to have a major effect on the fatty acid composition of algae (Lynch and Thompson, 1982; Murata et al., 1975; Renaud et al., 2002; Sato and Murata, 1980). As general, fatty acid unsaturation increasing with decreasing temperatures.

Table 4.1: Fatty Acids composition (%) of three different classes: saturated + monounsaturated (SFAs + MUFAs) and polyunsaturated (PUFAs)

% \bar{X} Fatty Acid		SFAs + MUFAs	PUFAs
Tubulars	0.33 d ⁻¹	61.87	29.28
	0.28 d ⁻¹	61.29	25.13
	0.12 d ⁻¹	73.16	19.75
Raceways	0.33 d ⁻¹	59.04	31.50
	0.28 d ⁻¹	49.56	39.75
	0.12 d ⁻¹	46.39	41.52
Flat Panels EW	0.33 d ⁻¹	47.03	41.97
	0.28 d ⁻¹	43.80	44.68
	0.12 d ⁻¹	40.85	47.39
Flat Panels NS	0.33 d ⁻¹	54.14	35.01
	0.28 d ⁻¹	49.25	42.90
	0.12 d ⁻¹	46.55	43.54

So, the possible biodiesel from raceway and flat panel in outdoor condition at this temperature shows a lower quality than the biodiesel from tubular, confirming this photobioreactor is more suitable for outdoor condition.

Conclusion

Microalgal biodiesel has been competitive with other biofuel and petroleum derived sources, for its industrial development. One of the major challenge is the biodiesel production from microalgae which has also a competitive cost. Chisti (2007) reported that in the United States during 2006, the petrodiesel prices ranged between \$0.66 and \$0.79 L⁻¹. This price includes taxes (20%), cost of crude oil (52%), refining expenses (19%) distribution and marketing (9%). If taxes and distribution are excluded, the real price of petrodiesel in 2006 was \$0.49 L⁻¹. Using the same analogy, a reasonable target price for microalgal oil is \$0.48 L⁻¹. So, require reduce the cost of production algal oil from \$2.80 L⁻¹ to \$0.48 L⁻¹. Tubular photobioreactors like shown in our study, is the most productive system but to be competitive their cost must be below the cost of the raceways systems. Acien et al. (2012) reported that tubular was most productive system, but it more expensive as compared to the raceways system. The authors also propose some technical aspects to decrease the cost, for example: (a) implementing extensive automatization of as many of the operations; (b) reduction cost of the equipment by a simplification of the design and the dimension of this latter when the production capacity is increased, (c) introducing a cheaper methods of sterilization like filtration with 1 µm membranes (d) reduced the cost of centrifugation by introducing flocculating-decantation system; (e) using flue gases from industrial sources can reduce the cost of carbon dioxide to value near to zero if flue gas are ready available; (f) the utilization of wastewaters containing mineral nutrients could reduced the cost for supplying of fertilizer for microalgal growth. Other methods for decrease the cost production were suggested by Chisti (2007) and Sánchez Mirón et al. (2003), in which a microalgal biodiesel can be converted in a biora refinery. In fact, biomass not contain only oil, but also a significant quantities of protein, carbohydrate and other nutrients. Therefore, the residual biomass from biodiesel production process can be used potentially as animal feed. Besides, some of the residual biomass may be used to produce methane by anaerobic digestion to generating the electrical power necessary for running the microalgal biomass production facility. Also genetic and metabolic engineering can also have a greatest impact on improving the economics of production of microalgal diesel (Dunahay et al., 1996; Roessler et al., 1994). One strategy can be the improving expression of enzyme that are involved in the pathways of fatty acids

synthesis (Zeng et al., 2011). For example in *Brassica napus* the over-expression of glycerol-3-phosphate dehydrogenase (G3PDH) resulted in 40% increased in the lipid content of seed (Vigeolas et al., 2007). the G3PDH catalyses the formation of glycerol-3-phosphate, and this suggests that genes involved in TAG assembly are important for total oil seed production. This result is further supported by several other studies in which over-expression of TAG assembly genes result in increasing seed oil contents (Derelle et al., 2006; Jako et al., 2001; Lardizabal et al., 2008). Another strategy for increased lipid accumulation is to decrease the lipid catabolism. In the lipid catabolism process, acyl-CoA oxidase, acyl-coA synthase, carnitine, acyl-transherase I and fatty acyl-CoA dehydrogenase are the key enzymes of β -oxidation of fatty acids. Research work focused on knocking out some of this enzyme genes has been reported to increased lipid storage (Derelle et al., 2006; Molnar et al., 2009). Several publications have shown that knocking out genes involved in *Saccharomyces cerevisiae* β -oxidation does not only lead to increased amount of intracellular fatty acids, but also extracellular fatty acids secretion (Michinaka et al., 2003; Scharnewski et al., 2008).

Several attempts have been made to improve the photosynthetic efficiency and reduce the effects of photoinhibition on microalgal growth. Most of this are focused on reducing the size of the chlorophyll antenna (Lee and Lee, 2003; Lewandowski et al., 2003). This approach may seem counterintuitive but it has two positive effects. It permits a higher light penetration in high-density cultures, and also allows a higher maximum rate of photosynthesis due to the fact that cells are less likely to subject photoinhibition since their light-harvesting complete absorbs less light (Melis, 2009). The present, showed the optimal dilution rate was at 0.28 d^{-1} in every photobioreactors and tubular shown the higher biomass volumetric productivity than the others photobioreactors. The biochemical data shown a higher content in lipid in tubular than the other photobioreactors, with a prevalence of saturated and monounsaturated fatty acids than the polyunsaturated fatty acid in almost all photobioreactors. In conclusion, this study shown a higher performance of tubular photobioreactors in terms of biomass productivity and fatty acids contents for biodiesel. No informations relative to the cost of this system are examined in this thesis even fought it is well known that they are very high and prohibitive to compete with other biodiesel resources like petroluem derived diesel.

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